

Computational Catalysis and Electrocatalysis

Perla B. Balbuena

Department of Chemical Engineering and
Materials Science and Engineering Program

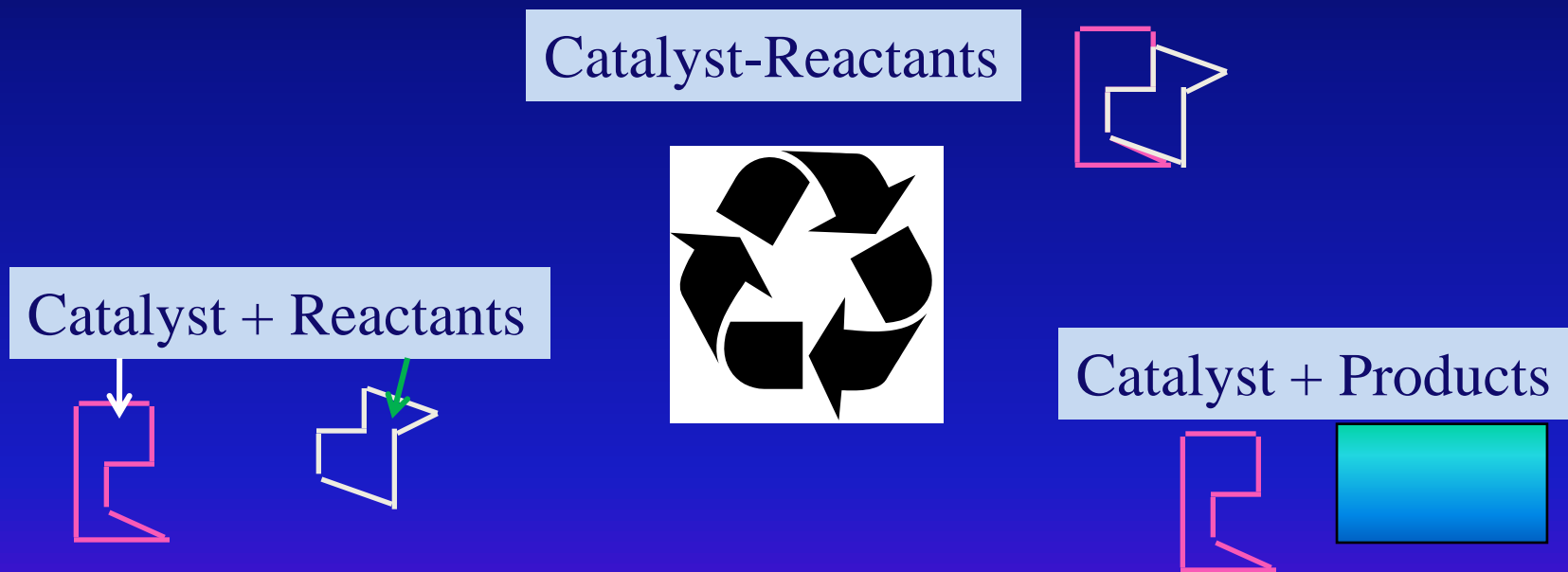
Texas A&M University

balbuena@tamu.edu

SC Annual User Meeting 2009, 20th Anniversary Celebration, May 6th, 2009

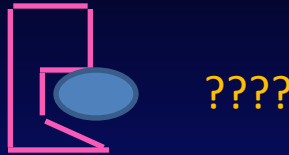
Why catalysis?

Most Reactions are too slow to be useful...



Catalysts speed up a chemical reaction
without being used up...

Why **computational** catalysis?



A good catalyst is a material whose surface is composed of active sites where reactants may be temporarily attached or may be decomposed

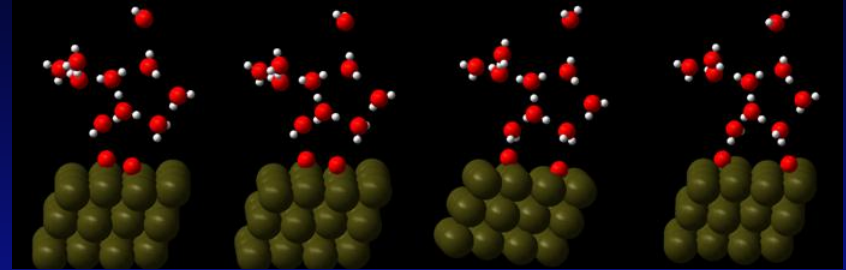
The challenge: find out an **efficient, durable, and cost effective** materials for catalysis

Experiments help, but... Too lengthy and expensive !!!

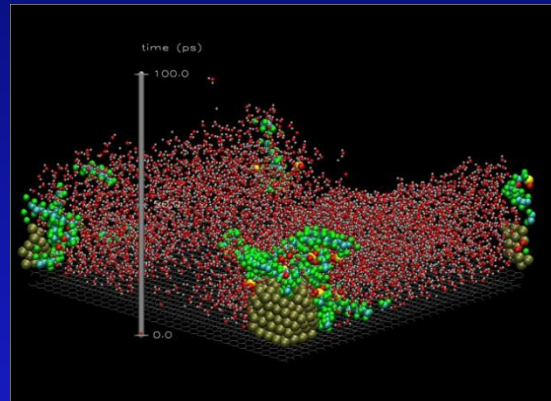
First-principles based computations are excellent tools to guide experiments and design novel materials

Tools-methods

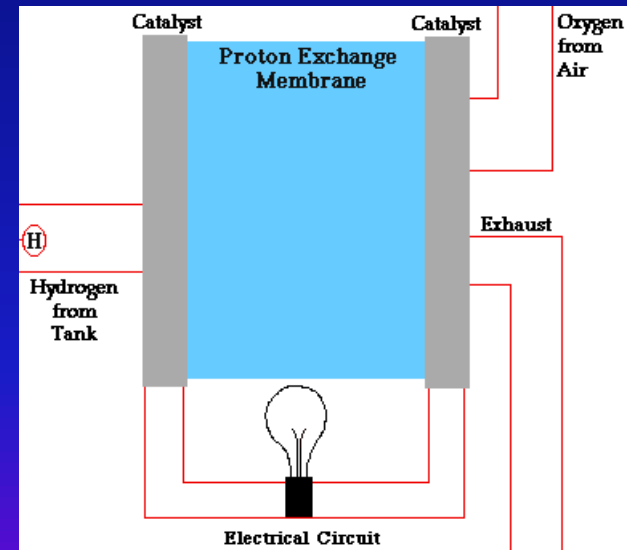
From the atomistic level...



To mesoscopic



And macroscopic systems...



Covering large time scales:
from femtoseconds to minutes, hours...

Tools-hardware

Novel catalytic designs
require reaching
the atomistic world

How?
Solving exact laws of
nature

Numerical solutions involving
realistic models now possible
because of supercomputers

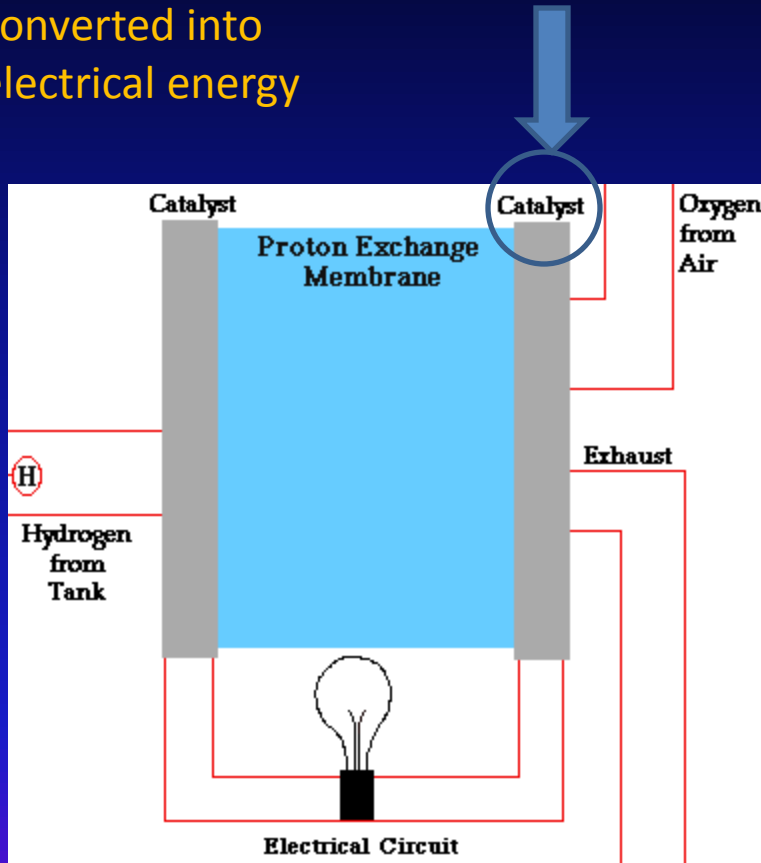


Applications from our research

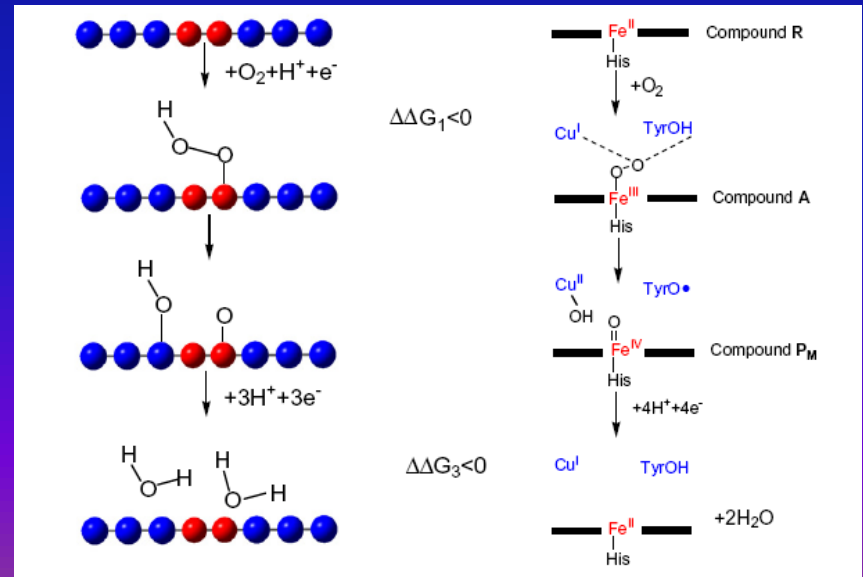
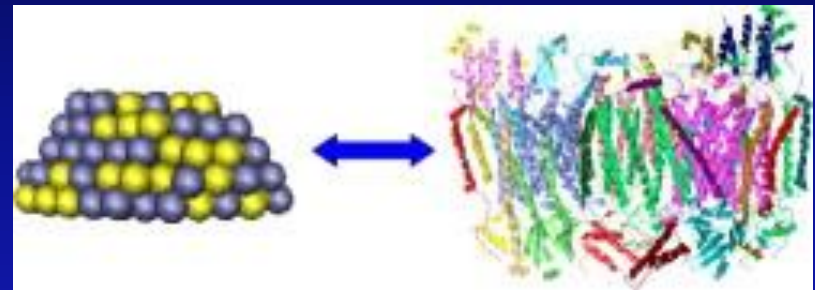
- Fuel cell electrocatalysts
- Controlled growth of carbon nanostructures
- Hydrogen storage
- Photocatalysis

Fuel cell electrocatalysts

Chemical energy is converted into electrical energy



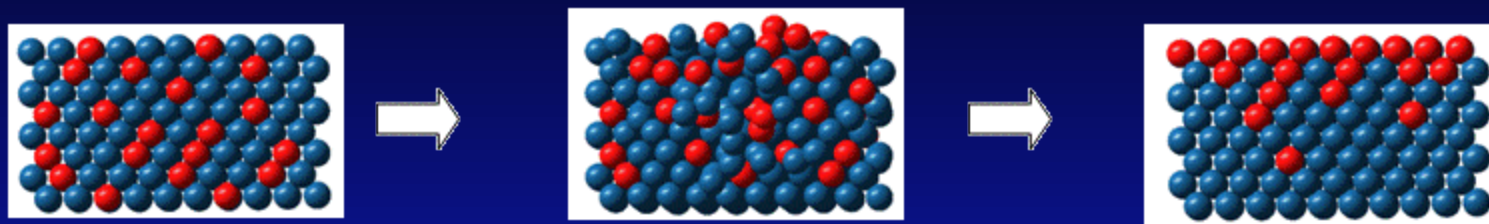
Parallelism between bimetallics and metalloenzymes



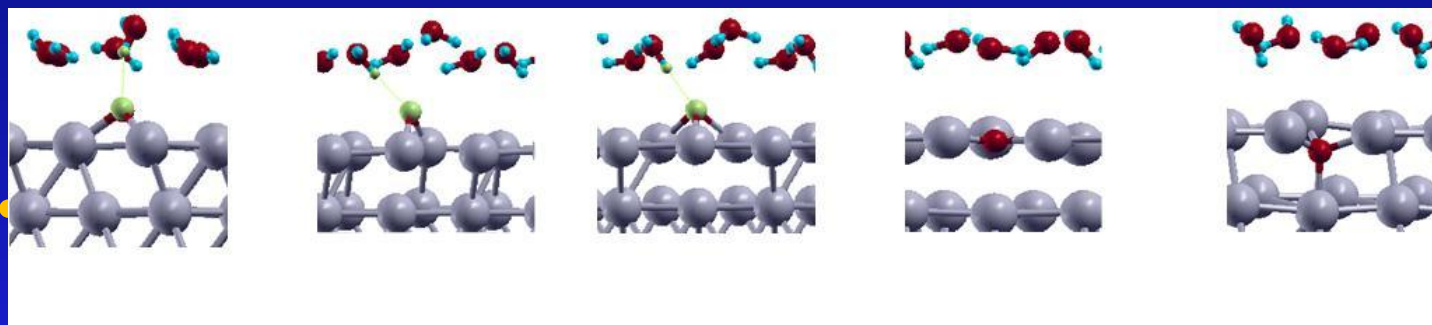
Wang and Balbuena, JPCB 2005;
Ma and Balbuena, CPL, 2007

Predictions and challenges

- catalyst surface evolution during reaction

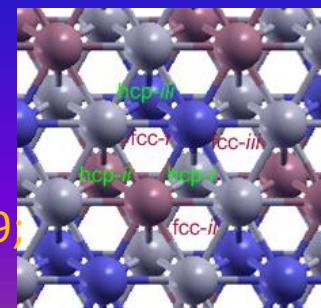


- catalyst degradation in acid medium



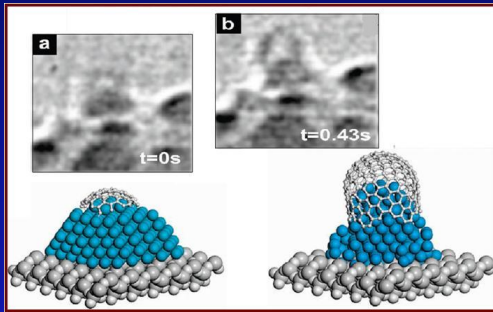
- new formulations of binary and ternary alloys

Ma and Balbuena, Surf. Sci, 2008; Ma and Balbuena, JPCC, 2008;
Ramirez-Caballero and Balbuena, CPL, 2008;
Callejas-Tovar and Balbuena, Surf. Sci, 2008; Hirunsit and Balbuena, Surf. Sci. 2009;
Martinez de La Hoz, Callejas-Tovar, and Balbuena, Mol. Sim., 2009



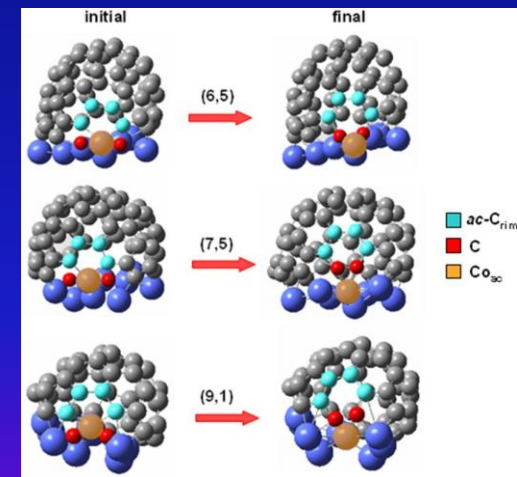
Controlled growth of carbon structures

Carbon structures (e.g. carbon nanotubes) grow over metal nanocatalysts at high temperatures



Robertson et al, Nanoletters, 2007

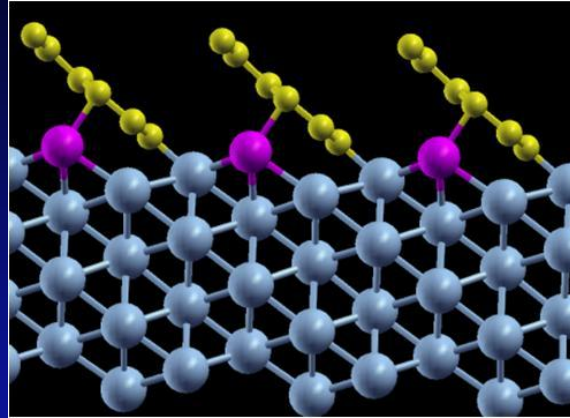
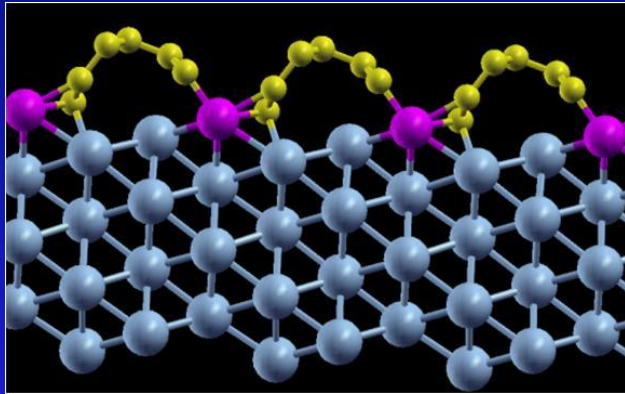
But a controlled growth is desired to form structures with specific properties



D. A. Gómez-Gualdrón and
P. B. Balbuena, Nanotechnology,
2009

Predictions and challenges

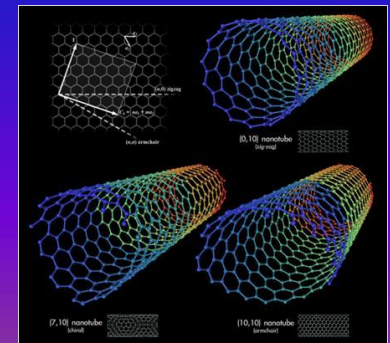
Graphene growth
parallel to the
(100) plane of Co



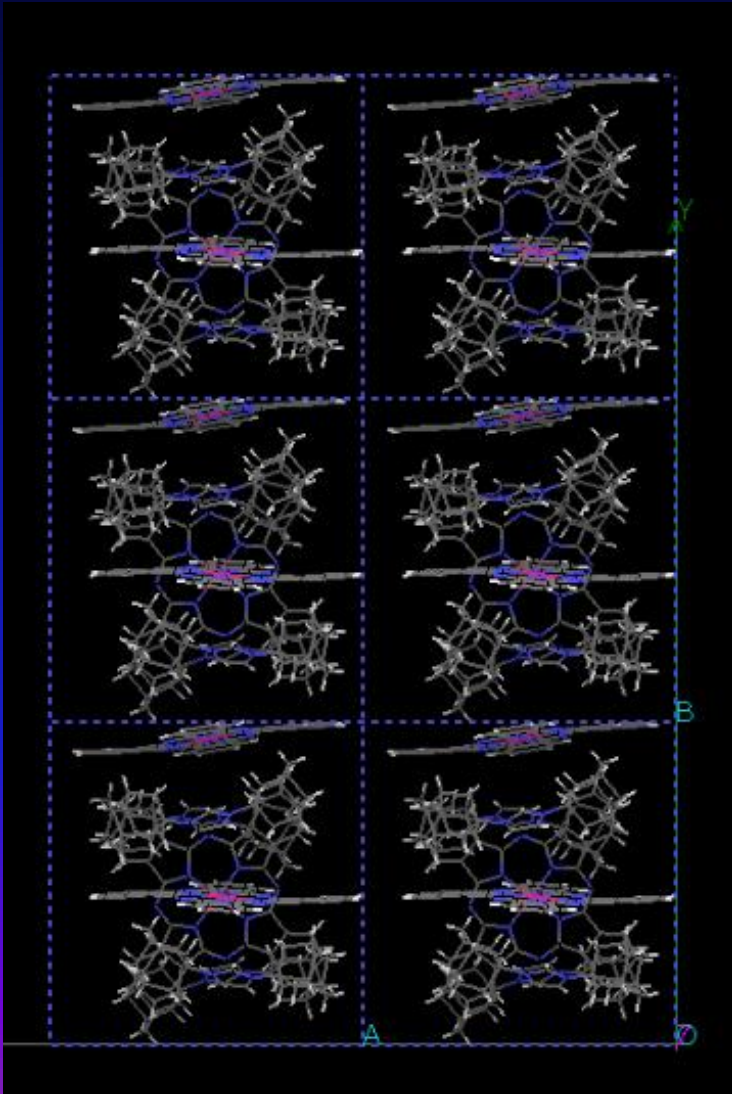
Formation of horizontally
aligned semi-nanotubes

G. E. Ramirez-Caballero, J. C. Burgos, and P. B. Balbuena
J. Phys. Chem. C (2009)

We are working towards predicting controlled nanotube helicity

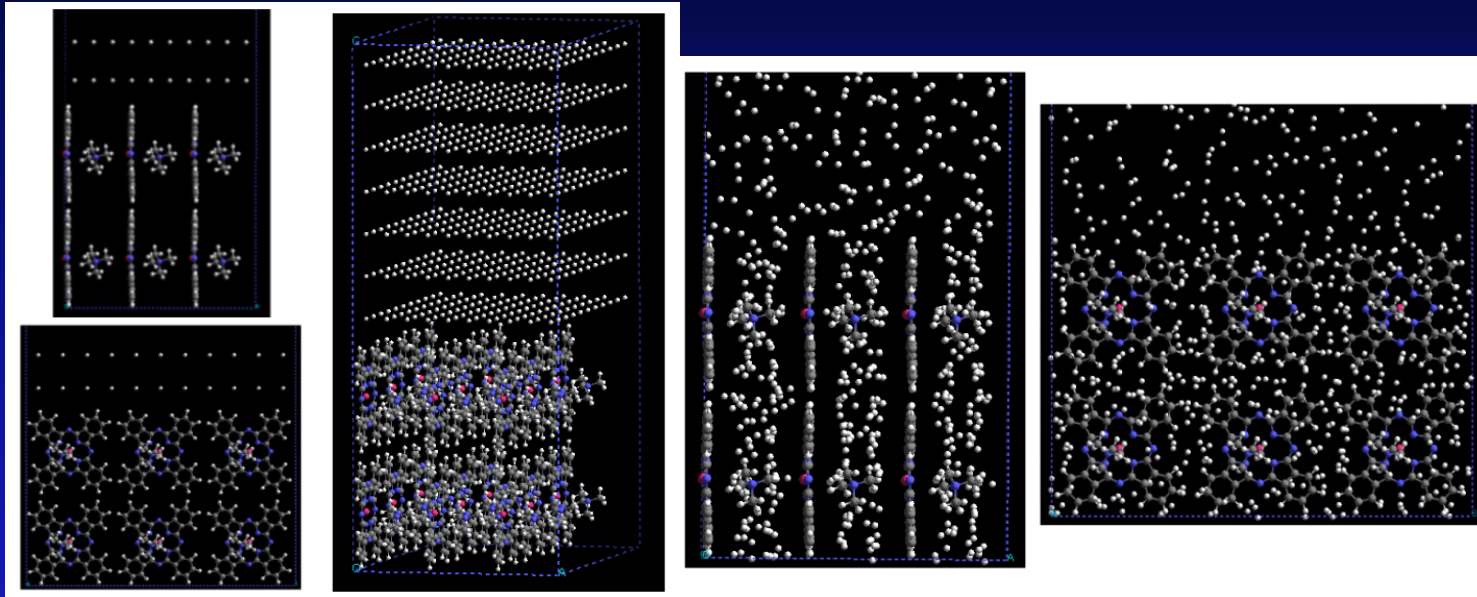


New materials for hydrogen storage



We are testing new materials that promise good ability for hydrogen storage—
Another fuel cell challenge

Predictions and challenges



Using molecular dynamics simulations we showed that certain layered materials have good storage capacity at moderate pressures and room temperature

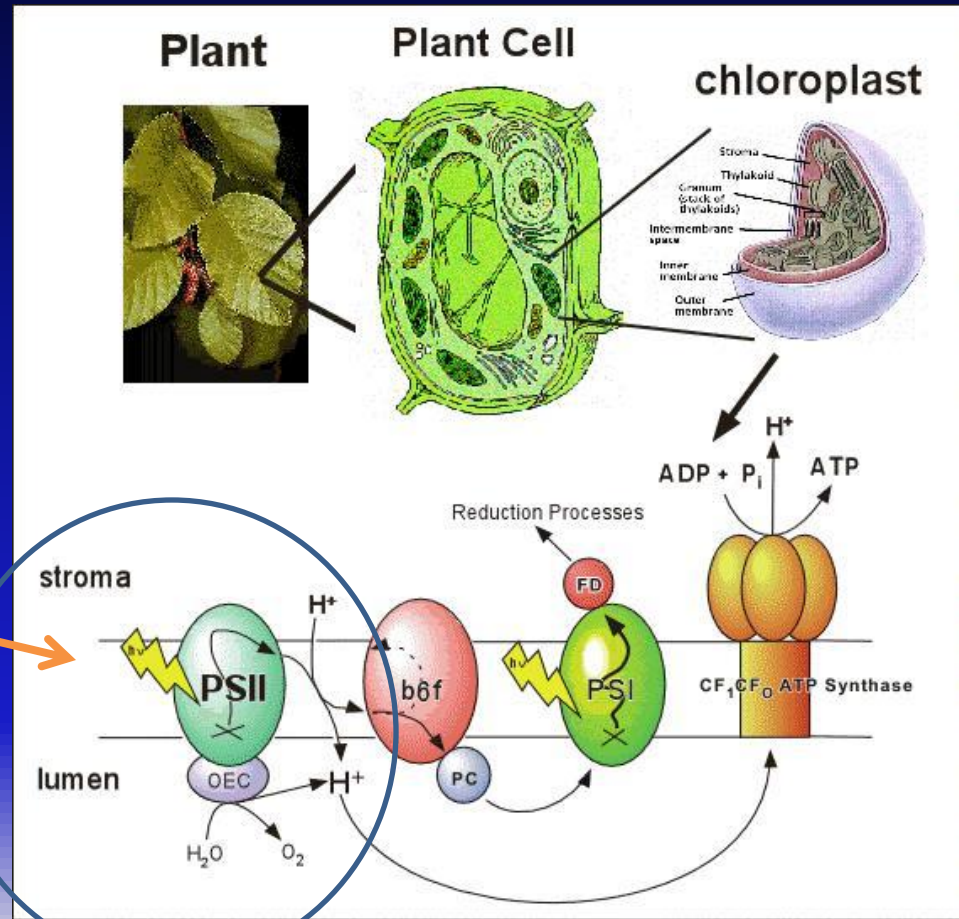
Lamonte, Gomez-Gualdron, Cabrales-Navarro, Scanlon, Sandi, Feld, and Balbuena, J. Phys. Chem. B, 2008

Future projects: photo-catalysis

Computational studies will be oriented to find new light-harvesting molecules and new catalysts for this important reaction

Photo-catalytic process

water decomposes over a catalyst producing oxygen and hydrogen



From members.tripod.com/beckysroom/pictures2.htm



Acknowledgements



Department of Energy/Basic Energy Sciences for financial support; grants DE-FG02-05ER15729; DE-FG02-06ER15836 and DE-FG36-07G017019

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