



# Characterization and Design of Nanomaterials Through Multiscale Modeling & Simulation

**Tahir ÇAĞIN**

Laboratory for Computational  
Engineering of Nanomaterials and Devices

<http://che.tamu.edu/orgs/groups/Cagin/>

Artie McFerrin Department of Chemical Engineering

Texas A&M University

E-mail : [cagin@che.tamu.edu](mailto:cagin@che.tamu.edu)

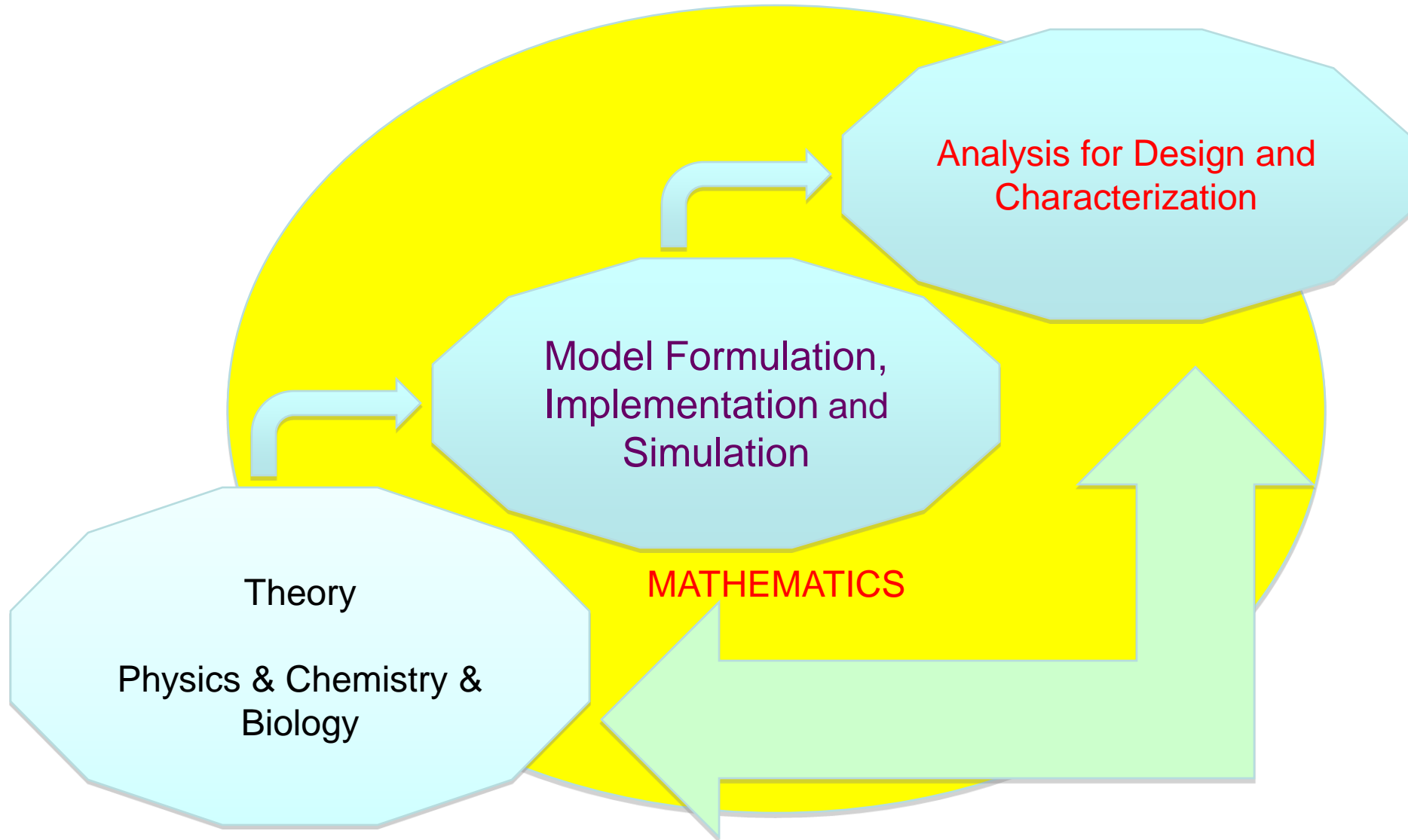


# Summary

- Rational Design and Characterization via Modeling and Simulation
- Multiscale Modeling Hierarchy and Materials Simulation
- Examples of Applications in Functional Materials
  - Multiscale modeling PMMA Thin Films
  - Hydrogen storage and Delivery (MOFs)
  - Thermo-electrics
  - Piezo-electrics and Ferro-electrics
  - Amplified Fluorescence Quenching Polymers for IED sensing
  - Structure, Assembly and Transport in Cyclic Peptide nanotubes (CPNT)
  - Stress Corrosion Cracking in Fe based alloys
  - Nuclear Fuel materials
  - Damage Cascade Simulations
  - Magnetic Shape Memory Alloys

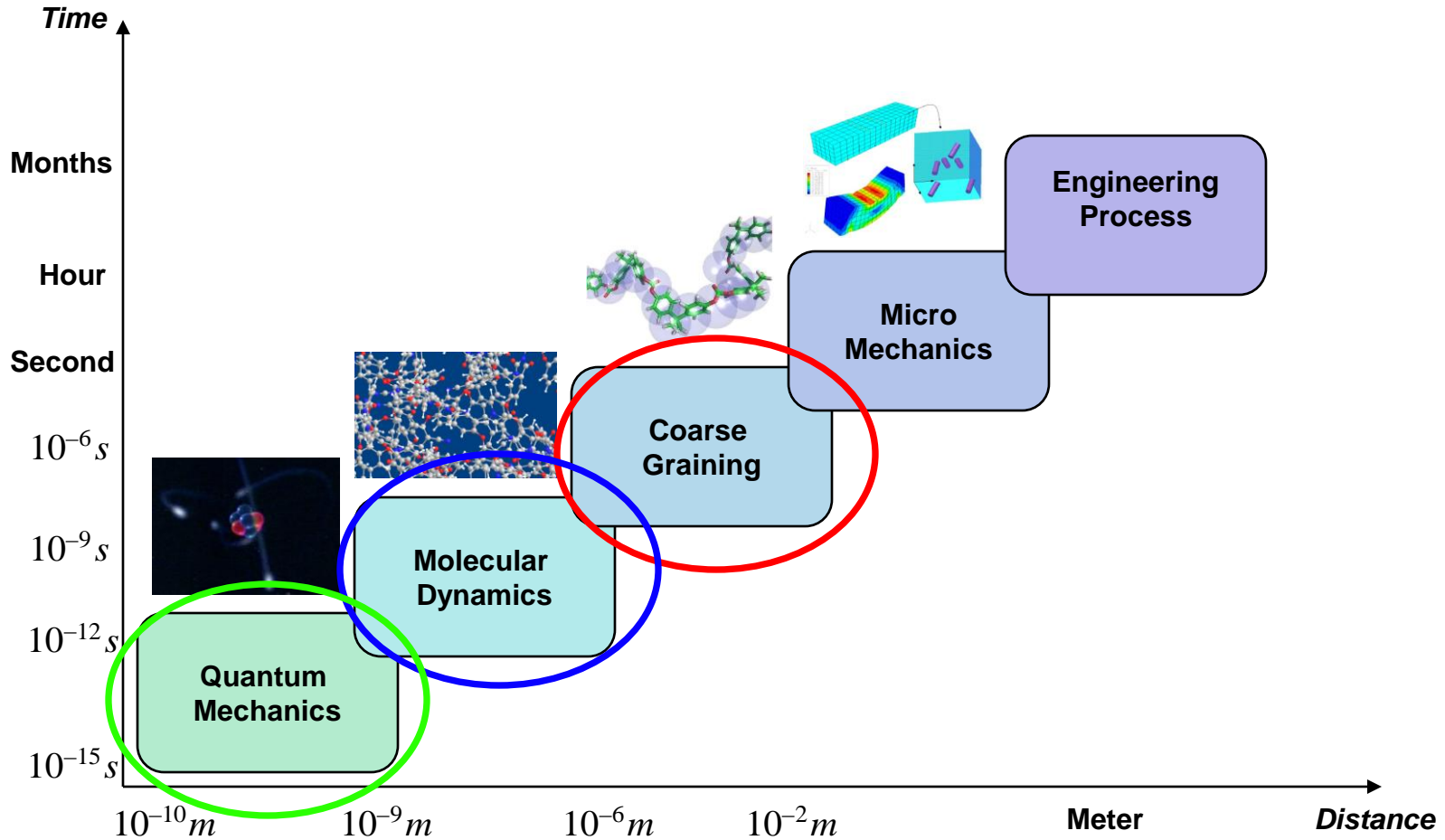


# Design/Characterization Through Modeling Paradigm





# Multiscale Simulation and Modeling Hierarchy





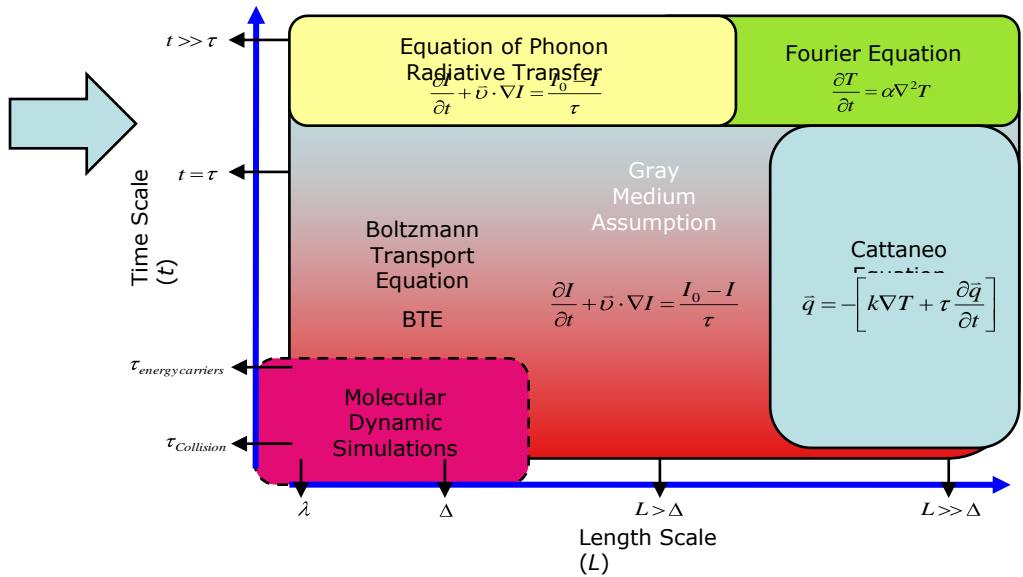
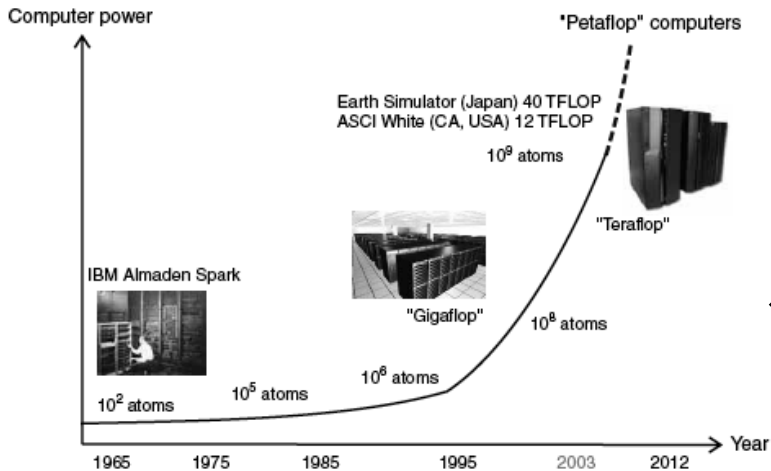
# Multiscale modeling of PMMA Thin Films for Microelectronics Applications



Develop a methodology for the numerical simulation of large models of polymeric thin films so that realistic estimation of mechanical and thermal properties can be obtained.

Diffusive equations, like the Fourier eq. for conduction heat transfer, do not account for the effect of energy carriers (i.e. phonons) in thermal properties.

Molecular Dynamic simulations can be implemented to solve this problem.



However, the computer resources needed to simulate a film of the necessary length and time scale (few tens of nanometers, microseconds) are prohibitive.

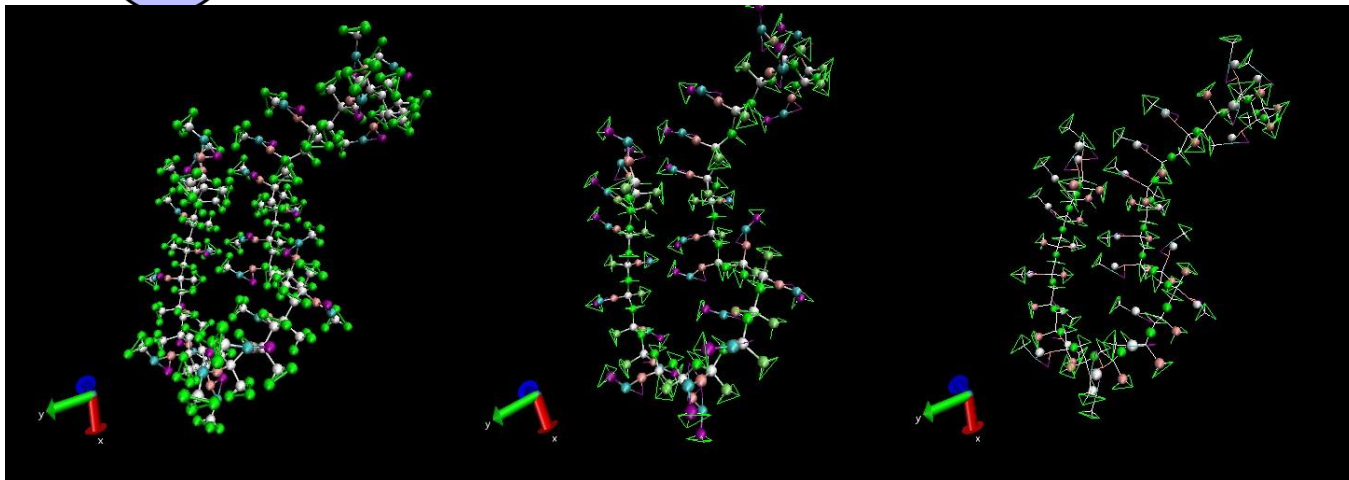
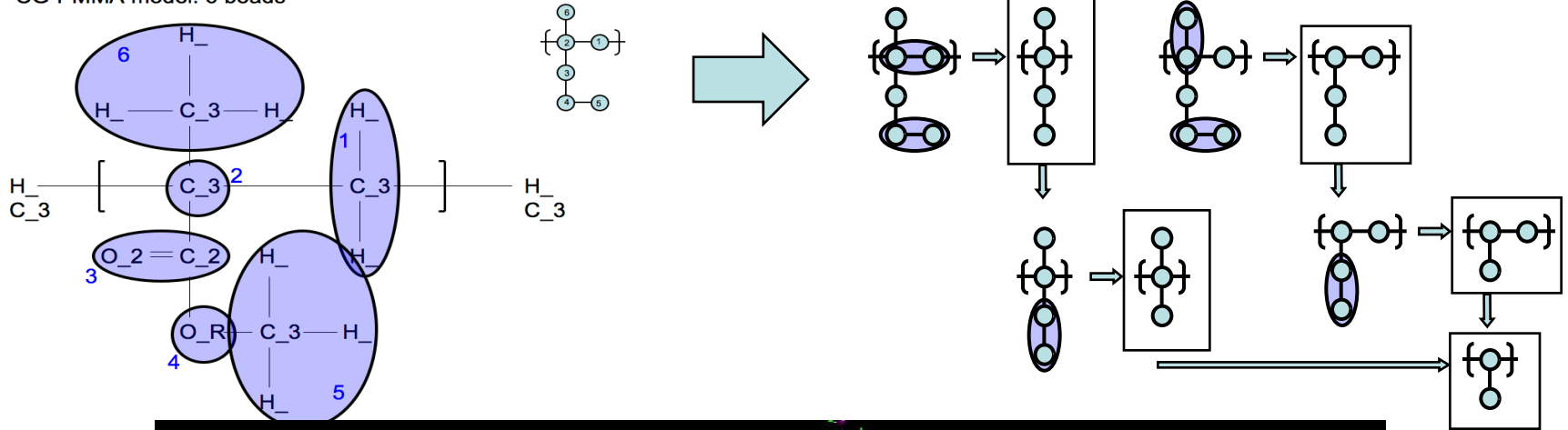


# Coarse Grain Molecular Dynamics



- Groups of atoms represented by a single bead
- Used for complex molecules in biosciences (proteins, DNA)
- Used in simulations of entangled polymer melts

CG-PMMA model: 6 beads





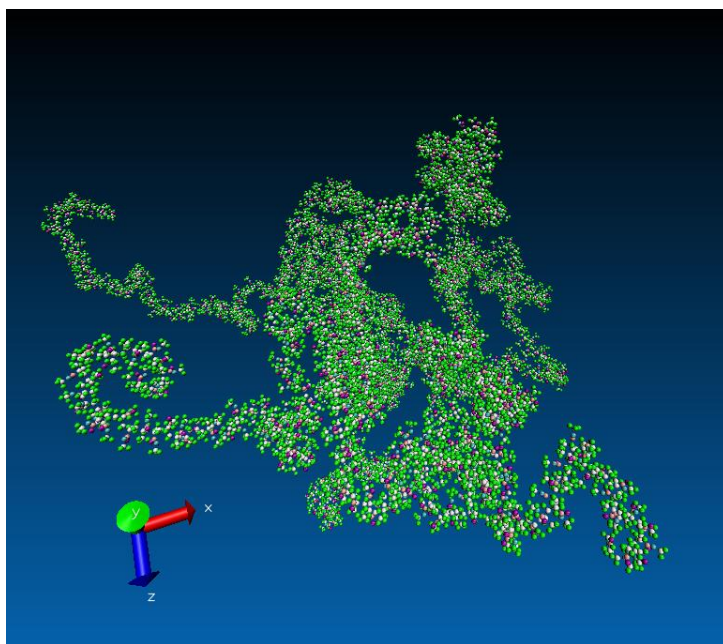
# Benefits



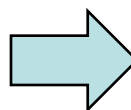
1 ns, local comp. (4-proc)

100 ns, Hydra (16-proc)

Model	Atoms/Beads	Run time (hrs)	Atoms/Beads	Run time (hrs)
Atomistic	2256	25.58	18002	~ 1000
6 Beads	900	4.85	7200	165.0
4 Beads a	600	2.05	4800	46.6
4 Beads b	600	2.45	X	
3 Beads a	450	1.28	3600	21.1
3 Beads b	450	1.90	X	
2 Beads	300	0.85	X	



A system of 18k atoms of PMMA is a 5x5x5 nm box.



Few tens of nanometers and microseconds are now attainable

A 20x5x5 nm box of 3-bead model running for 1  $\mu$ s on Hydra (64-proc) will take ~ 14 days.

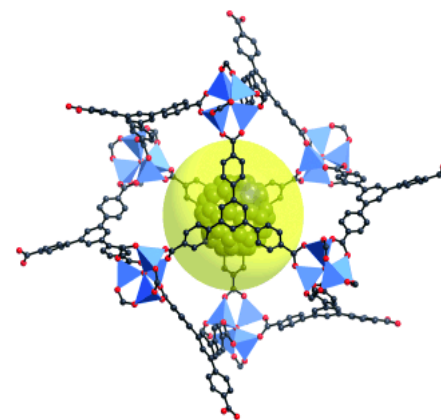
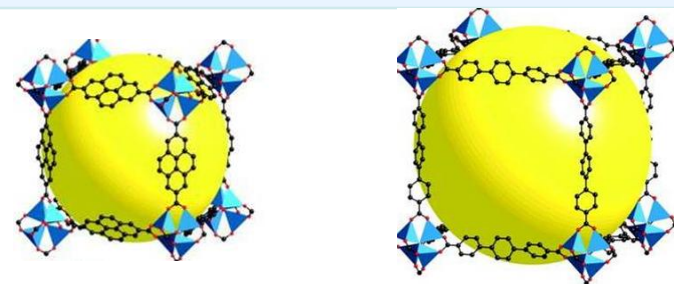
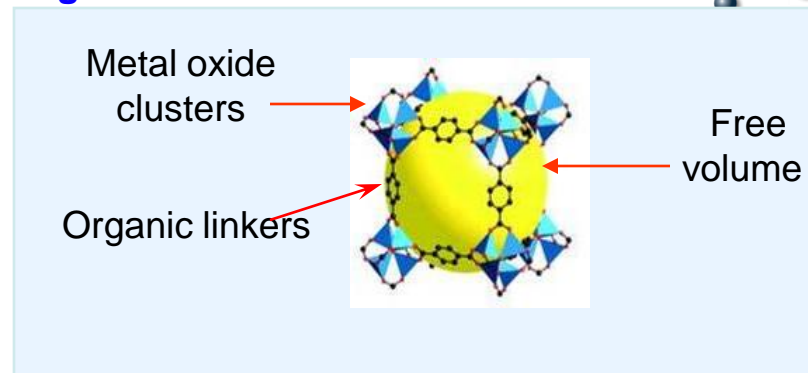


# Metal Organic Framework (MOF) for High Capacity Hydrogen Storage and Delivery

M. Mani Biswas, T. Cagin



- Crystalline material
  - Metal oxide clusters at vertexes,
  - Connected by organic linkers.
- Porous, large surface area (2500 - 5000 m<sup>2</sup>/gm)
- Low density (0.59 gm/cc)
  
- Crystals can be designed
  - Geometry, pore size can be varied (3.8 - 30 Å)
  - Linker molecule of different chemistry can be chosen
  
- Selective storage of guest molecule inside free volume
  - Hydrogen gas storage, gas separation
  - Drug delivery vehicle
- Designable property
  - Catalysis, molecular detection

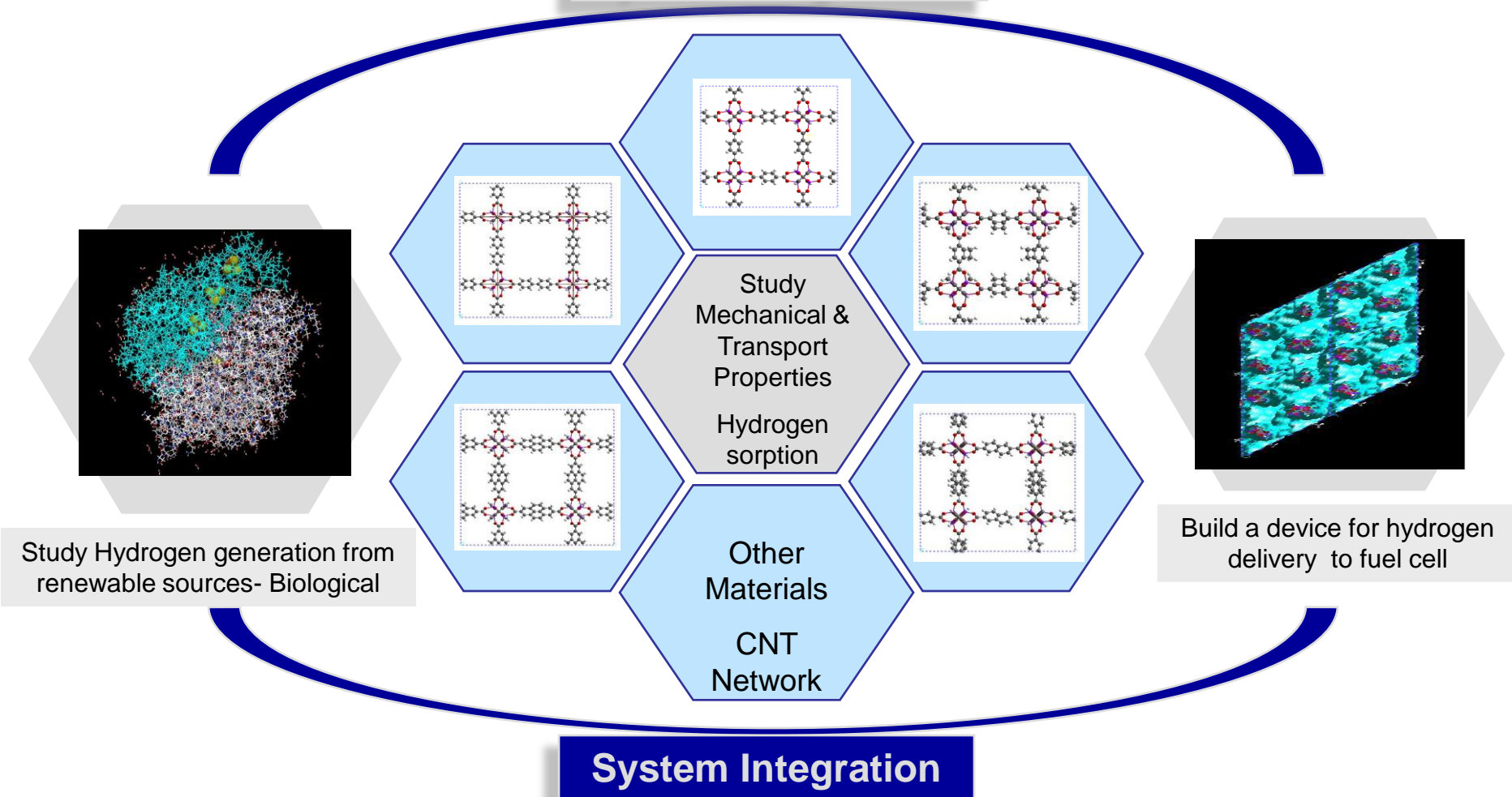


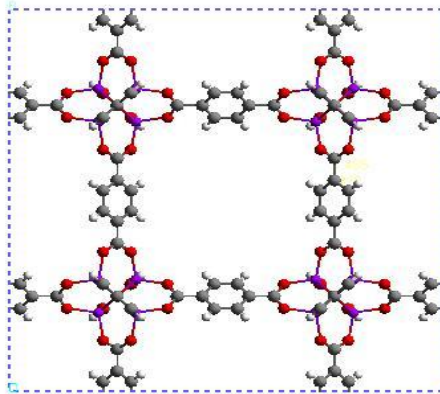




Theoretical Investigation using Classical MD simulations and Quantum Level calculation  
- properties of Metal Organic Frameworks (MOF) for efficient hydrogen storage and delivery

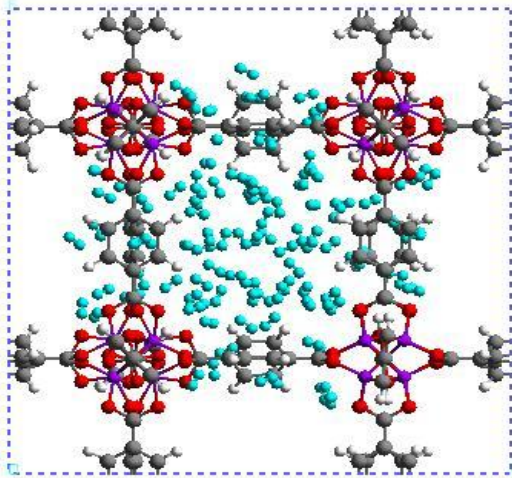
## System Integration



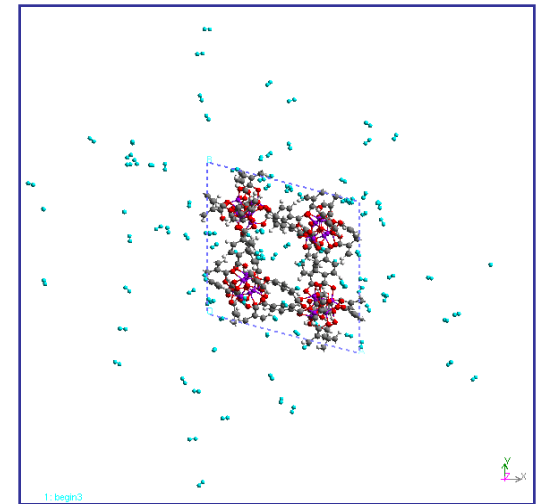


Loading at 100 MPa (298k)

Depressurize at 100 MPa



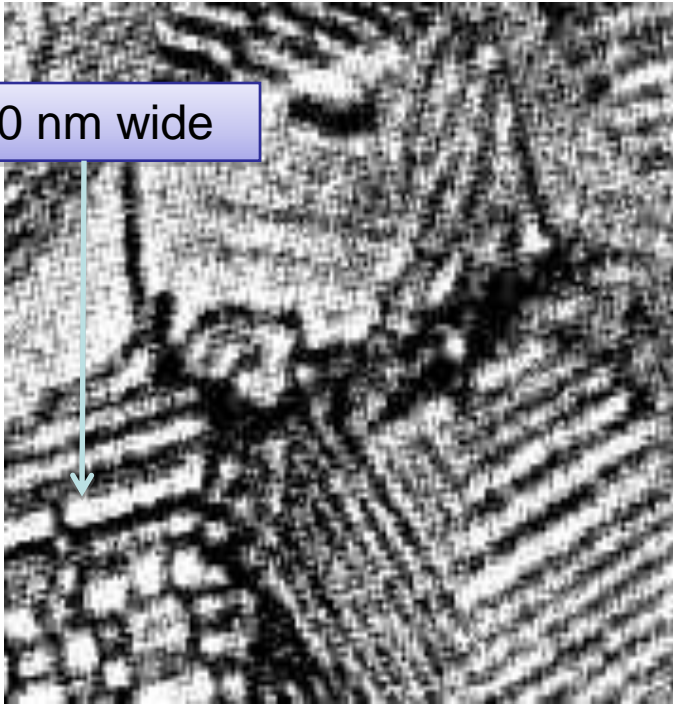
Unloading at 300 MPa (298k)



M. Mani-Biswas, T. Cagin, "Shape memory effect in MOFs", to be submitted.

Domain Wall: Interface of polarization domains

1-10 nm wide

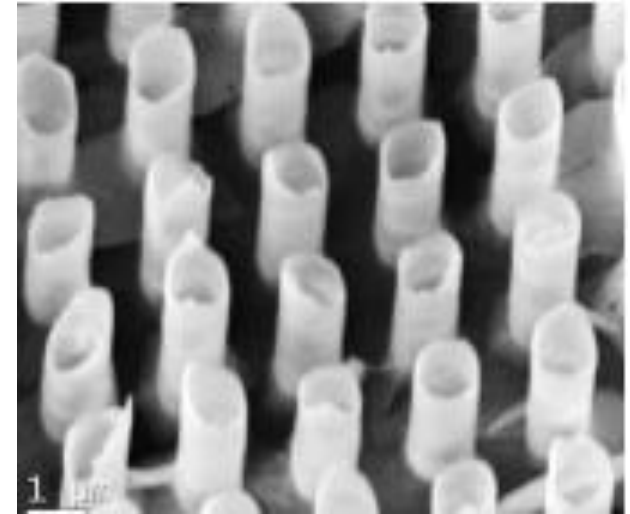
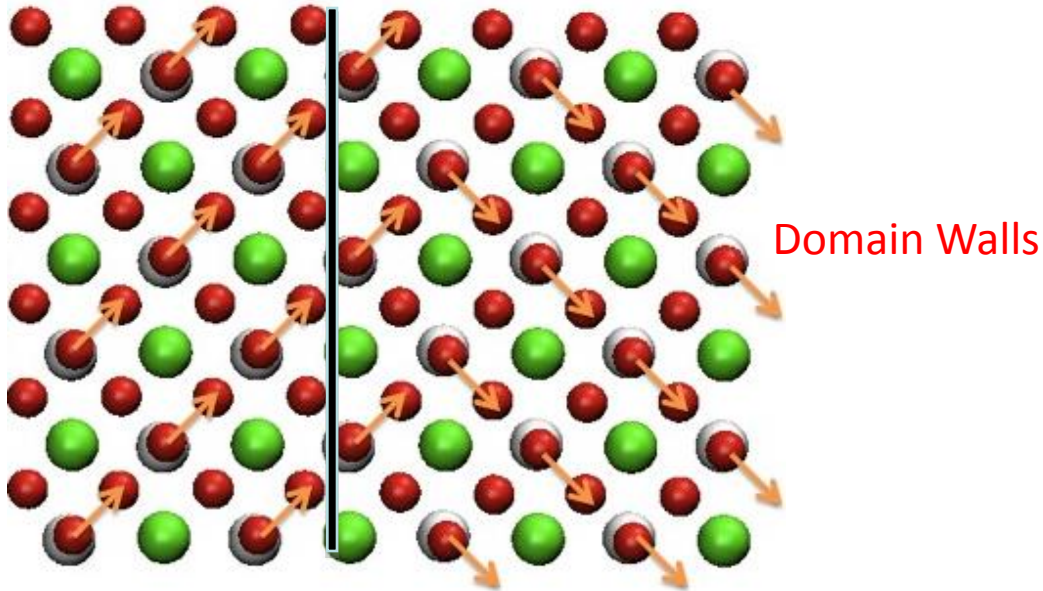


<http://www.materials.leeds.ac.uk/luec/actMats/Domain2.jpg>

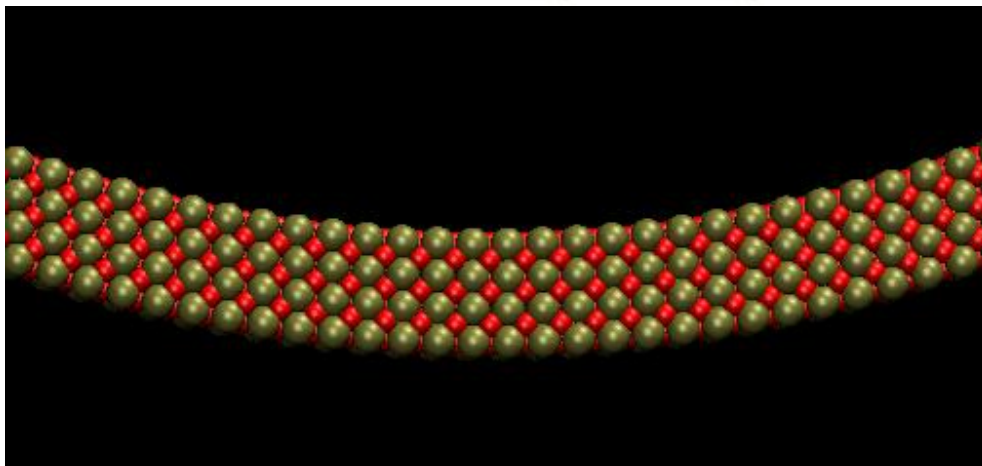
- Determine piezoelectric response and macroscopic polarization
- Fatigue switchable polarization
- Used in many applications
  - RAM
  - Actuators
  - Transducers
  - Sensors

Zhang, Cagin, Goddard, PNAS 103, 14695 (2006); Cagin et al, CMES 24, 215 (2008); Majdoub, Sharma, Cagin, PRB 78, 12407 (2008); PRB 77, 125424 (2008)  
J. Haskins, A. Kinaci, T. Cagin in progress.

Simulations excel in investigating nanostructures and the origin of bulk properties.



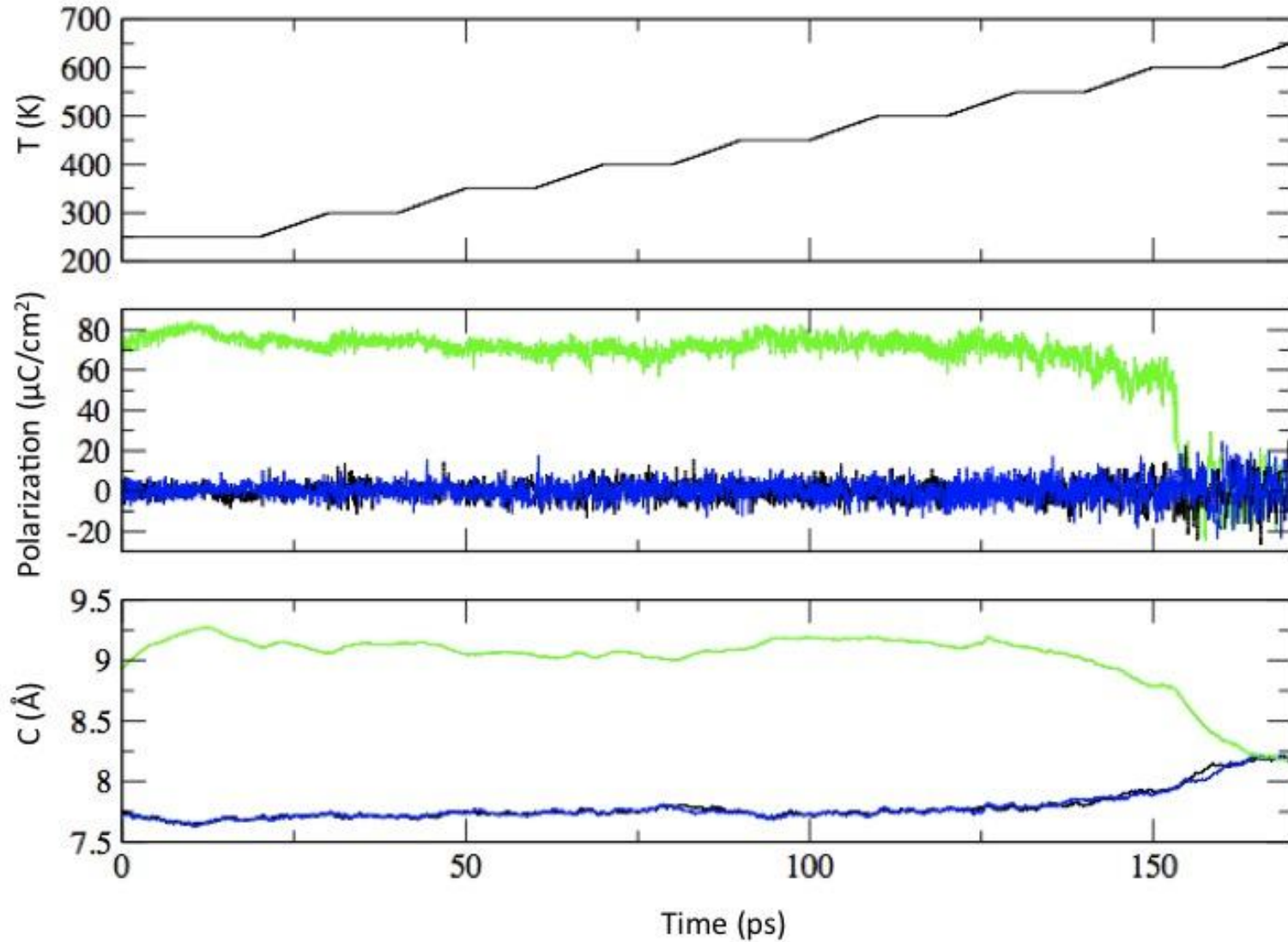
PZT nanotubes for memory devices



Nonlinearly strained cantilever polarization enhancement

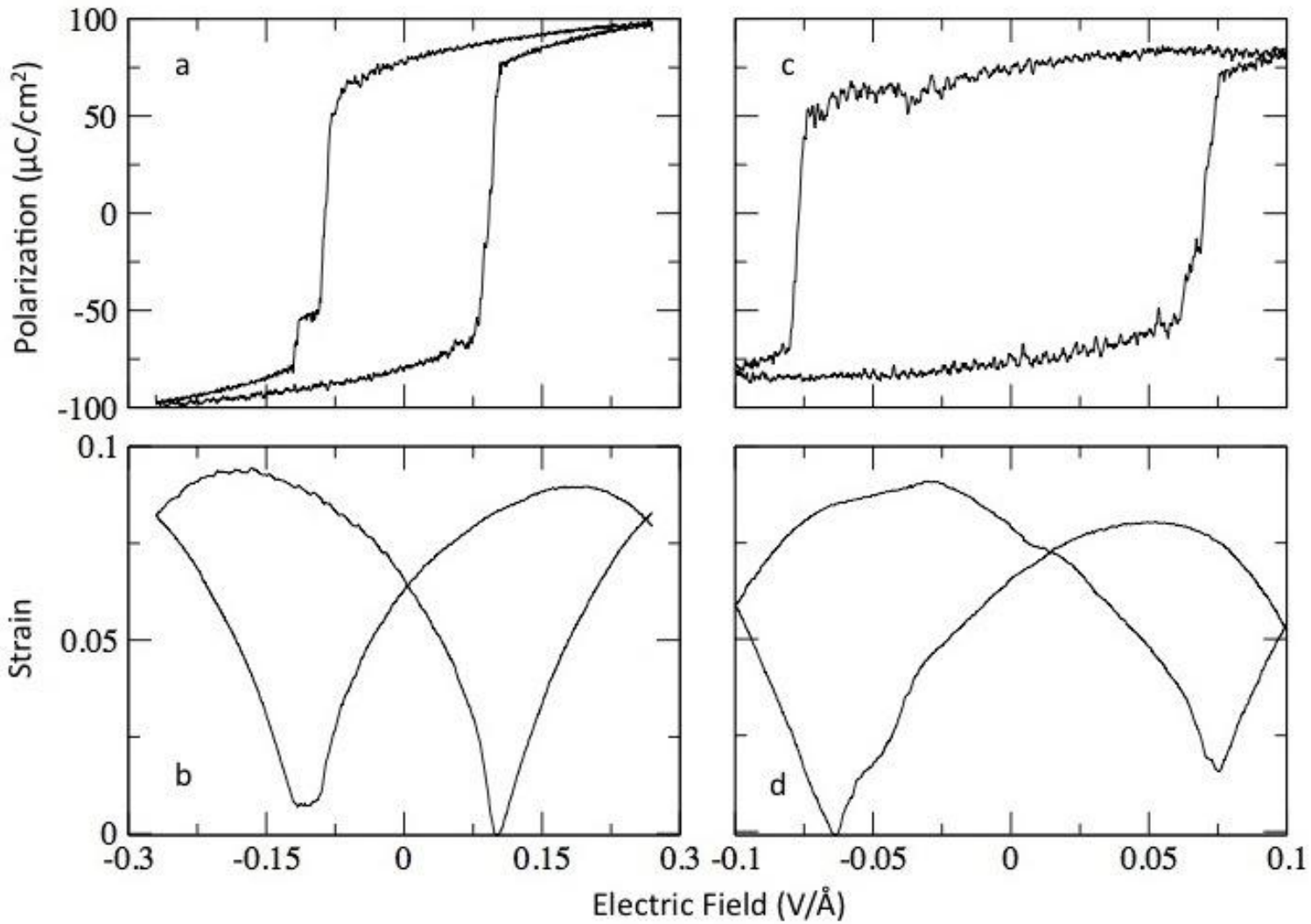


Temperature behavior of PZT calculated by polarizable force fields.





# Hysteresis behavior of PT and PZT.



Triangle field of 1 GHz with maximum strength of 0.27  $\text{V}/\text{\AA}$  .

The simulations shows characteristic ferroelectric hysteresis behavior.



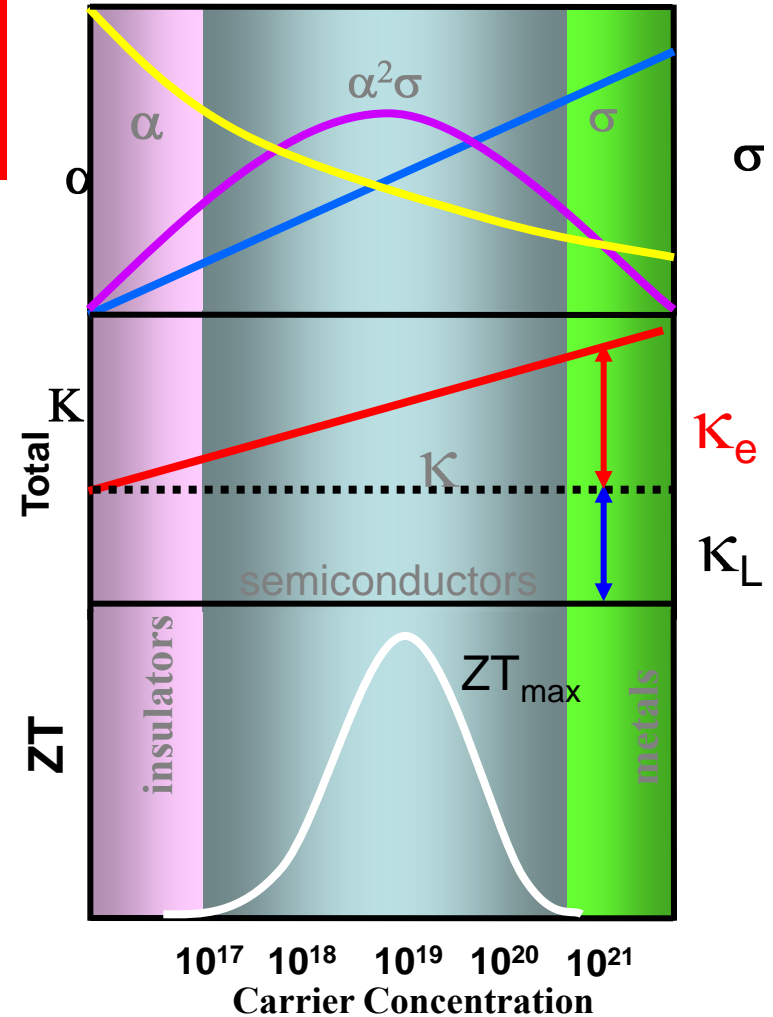
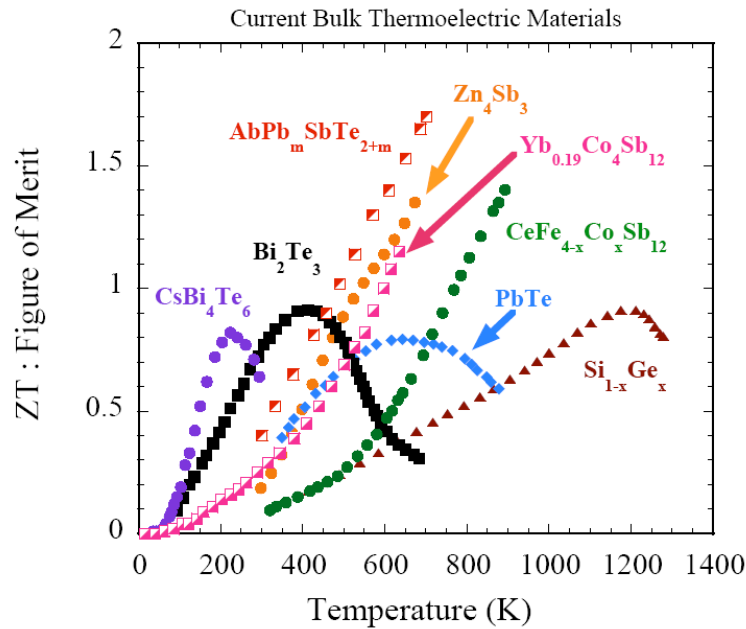
# ThermoElectrics, Performance Criteria: *Figure of Merit*



**Problem** : Inter-dependence of  $\sigma$ ,  $\kappa$  and  $S$  through carrier concentration.

$$ZT = \frac{S^2 \sigma}{\kappa} T$$

Increasing ZT is difficult - conflicting Properties



GF Wang and T. Cagin, Appl. Phys. Lett. 89, (2006) 152101

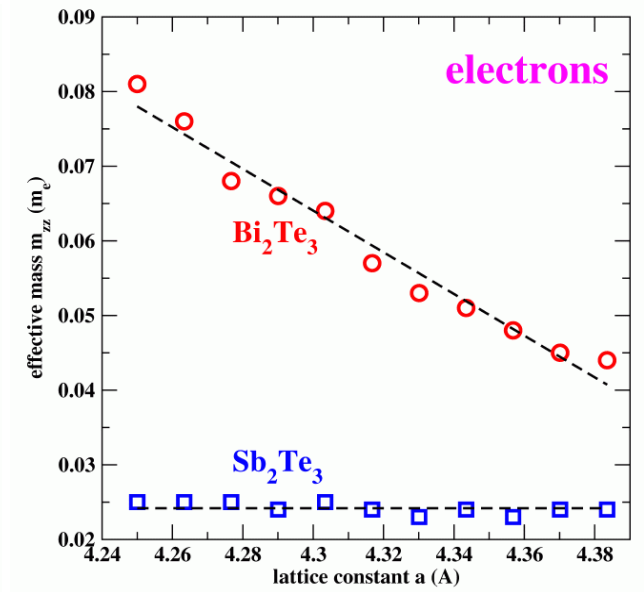
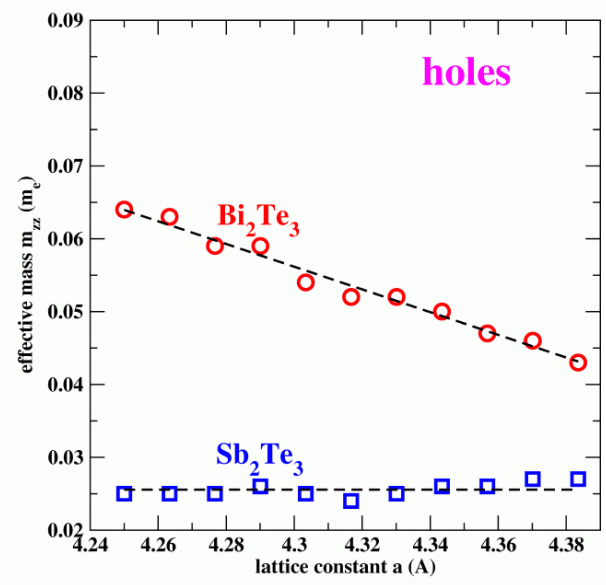
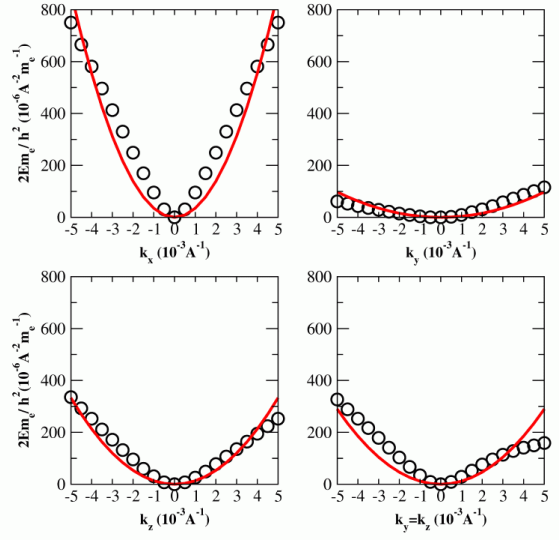
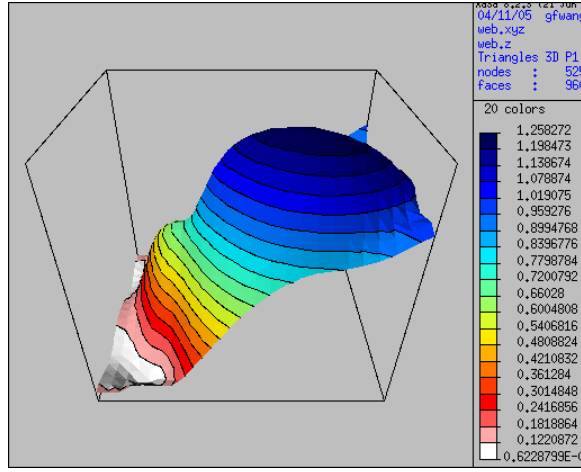
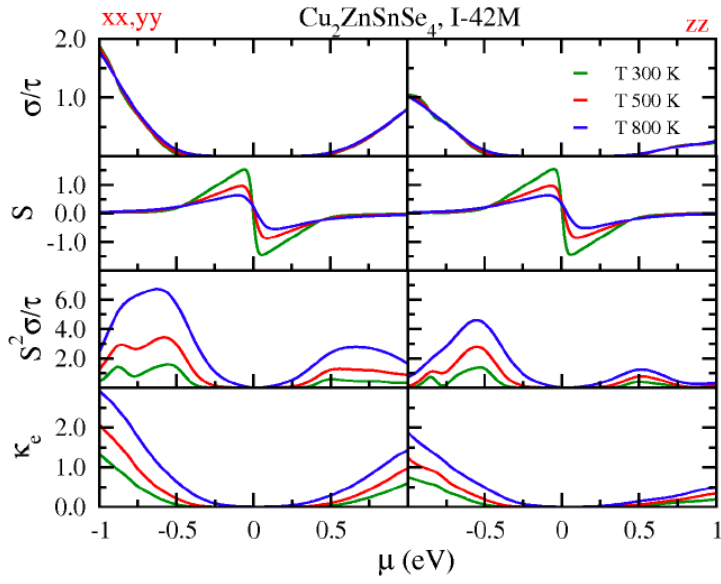
GF Wang and T. Cagin, Phys. Rev. B 75 (2007) 075201

C. Sevik, T. Cagin, in progress

A. Kinaci, C. Sevik, T. Cagin, in progress



# Transport Properties From Ab initio Theory



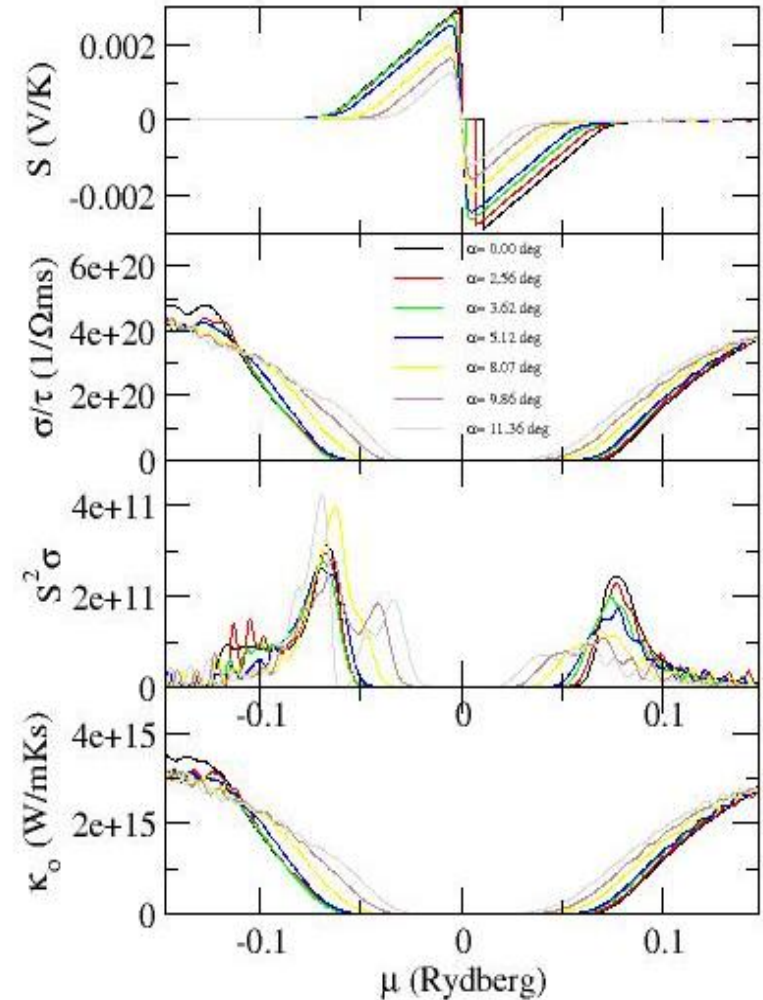
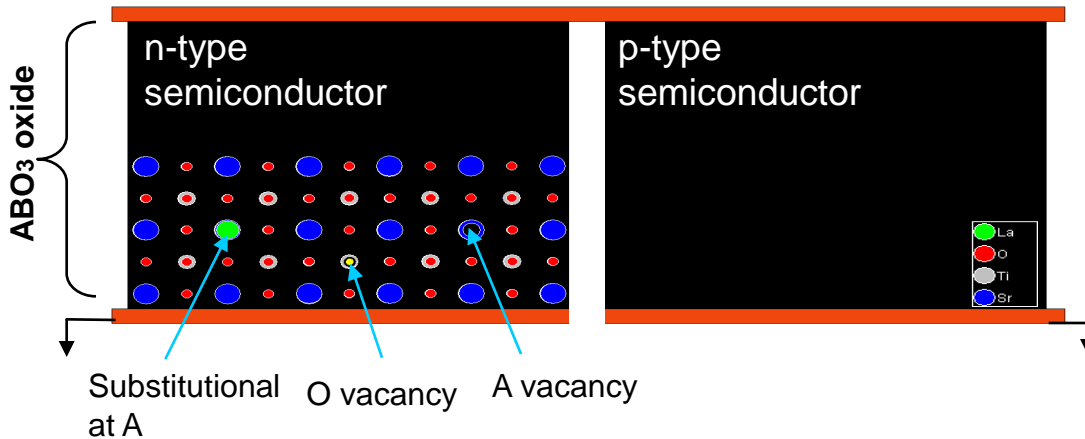




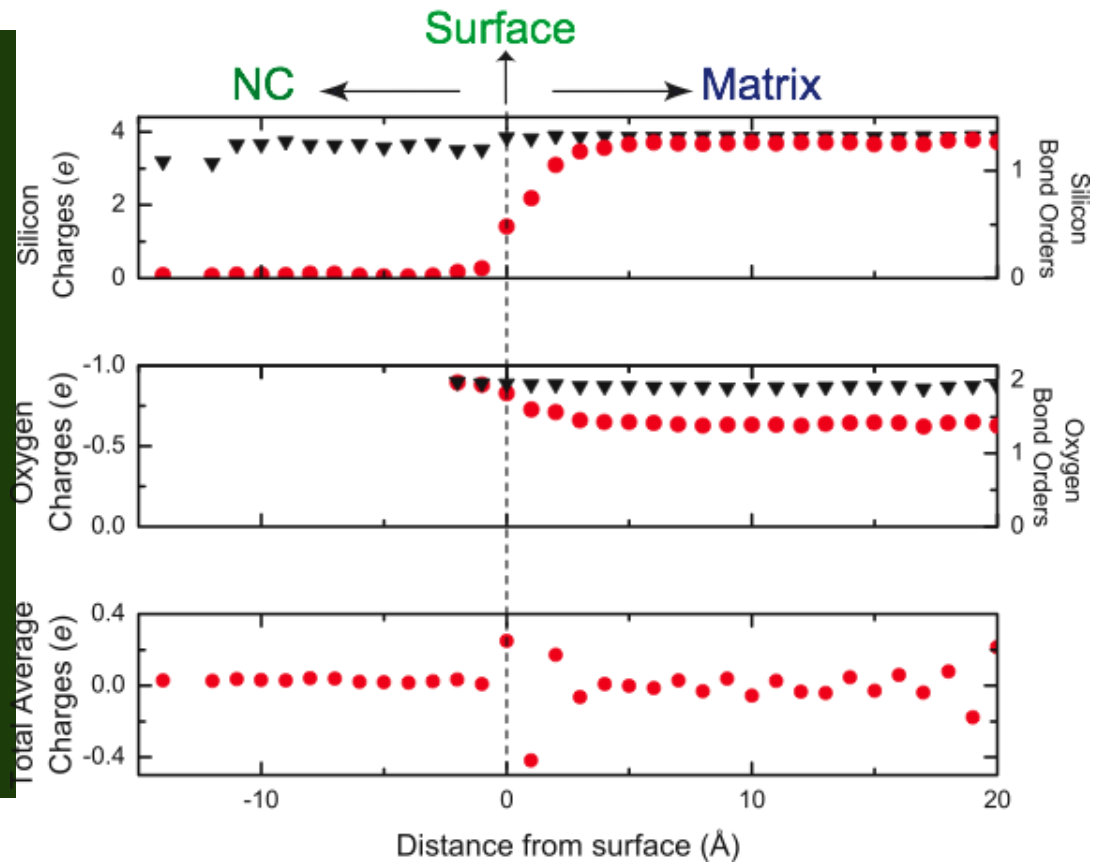
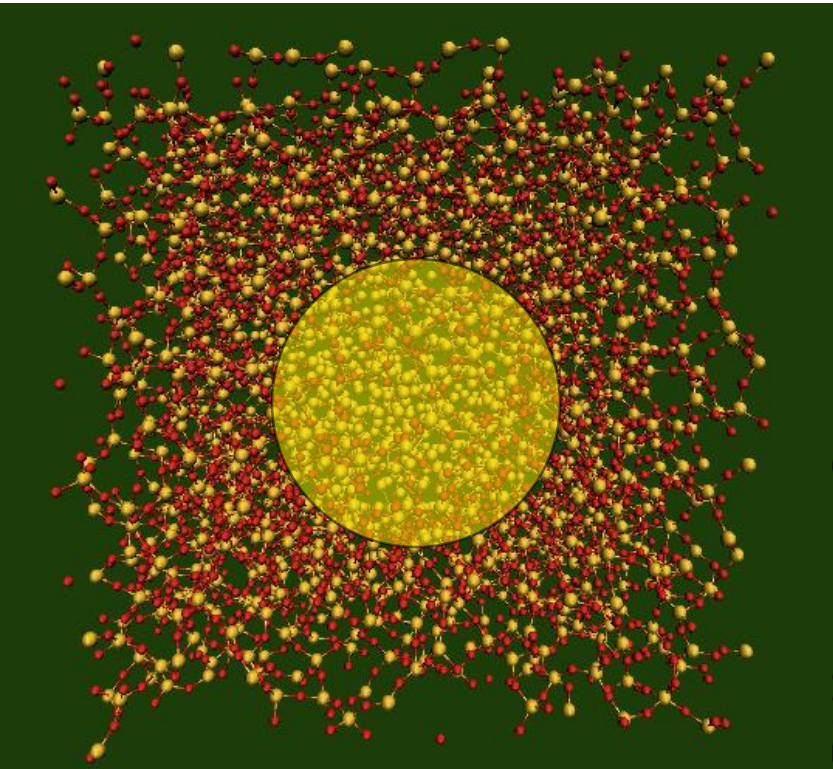
➤ New trends in thermoelectrics: Complex oxides and structural miniaturization (superlattices, nanowire, quantumdots ...)

## Manipulating properties of SrTiO<sub>3</sub>

- External stress
- Chemical alloying
- Controlled defects
- Structuring in atomic scale etc...



**Effect of simple shear on conduction properties of SrTiO<sub>3</sub>**

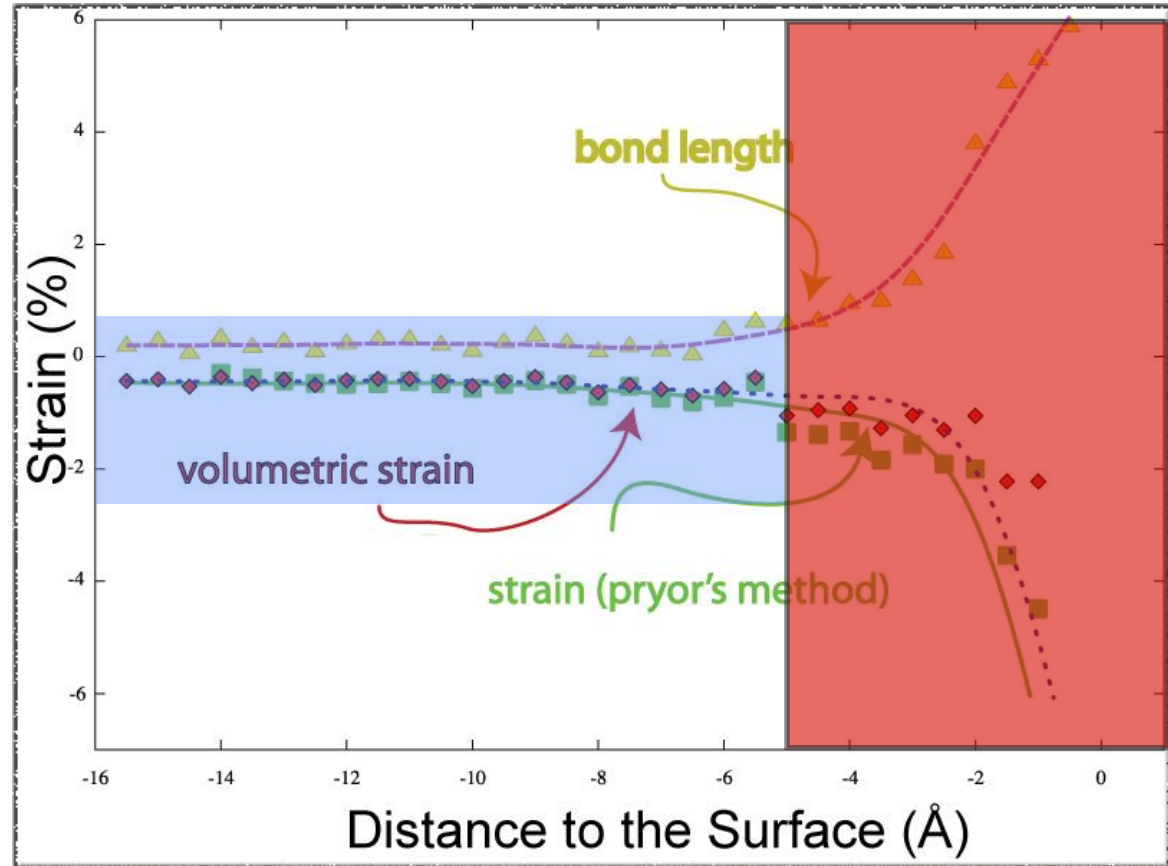


D. Yilmaz, C. Bulutay, T. Cagin, Phys. Rev B 77, 155306 (2008)  
D. Yilmaz, C. Bulutay, T. Cagin, Appl. Phys. Lett., (2009) in press



# Strain Profile in Si Nanocrystals

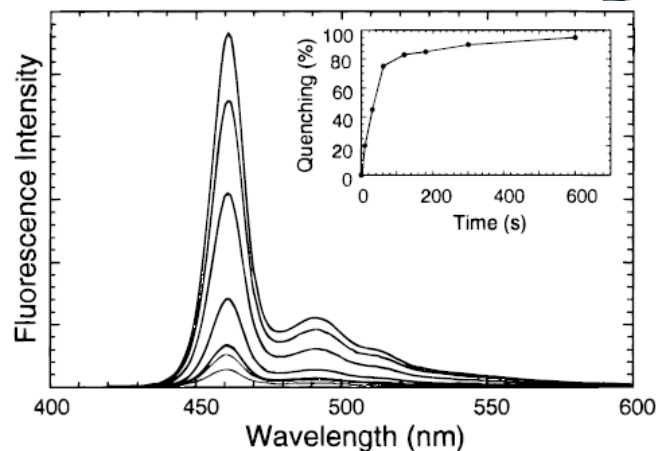
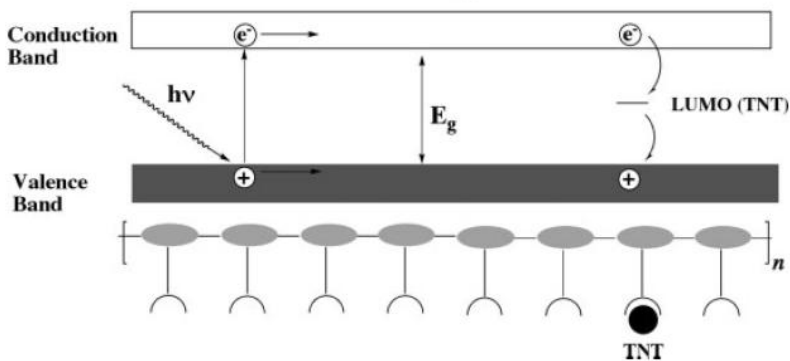
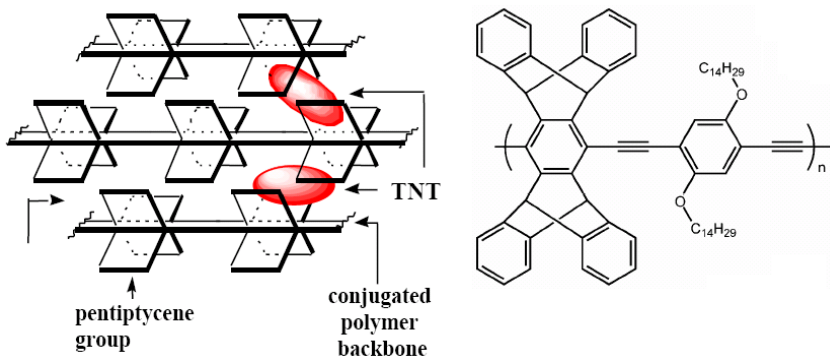
- ❖ Core of the NC is unstrained.
- ❖ Volumetric strain and hydrostatic strain (calculated with Pryor's method) shows similar behavior as expected.
- ❖ Bond lengths and strain shows opposite behavior.



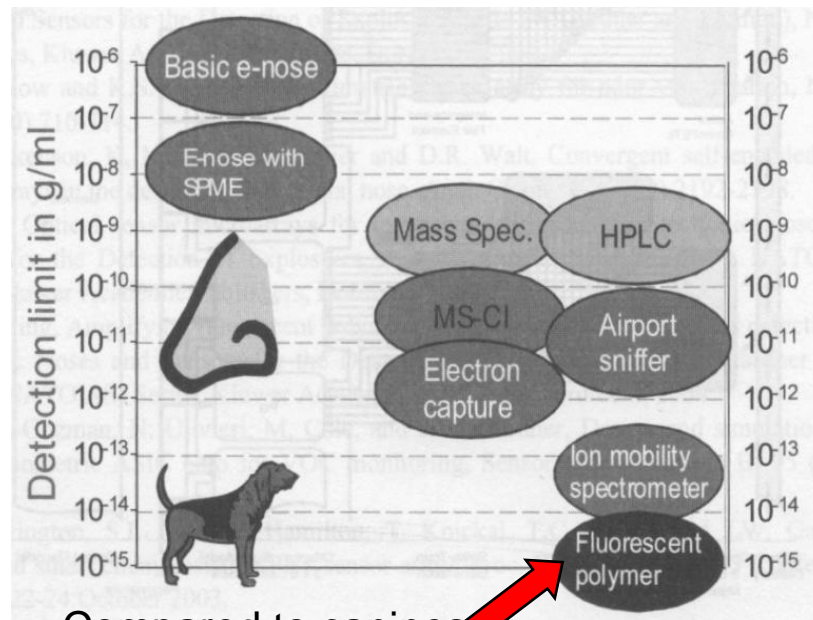


# TNT-AFP complexation

## Amplifying Fluorescent Polymer



$10^{-15}$  g detection limits (Fido™, Nomadics Inc.)



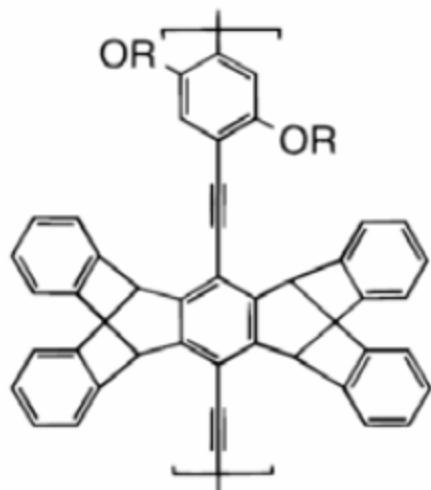
Compared to canines

Yang, J.S. and T.M. Swager, JACS 120, 1998.

B. Arman, H. Fan, T. Cagin, **Quantum Chemical Study of Sensing Mechanism of Nitroaromatics by Amplified Fluorescent Quenching Polymers** J. Chem. Phys. submitted.

# Molecular Wire Concept

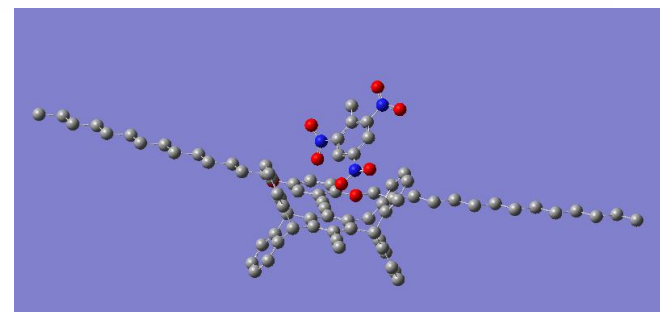
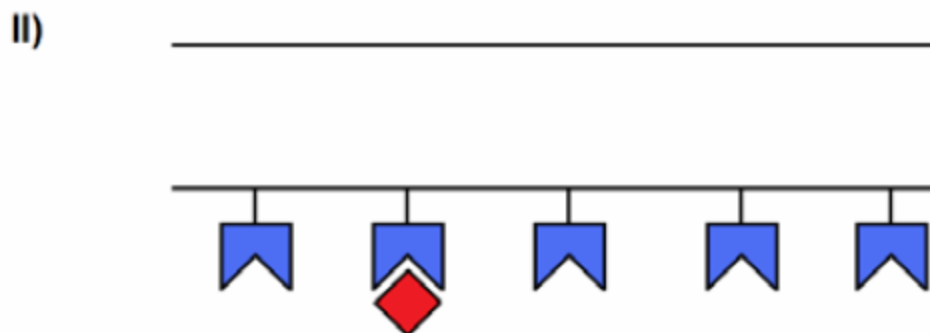
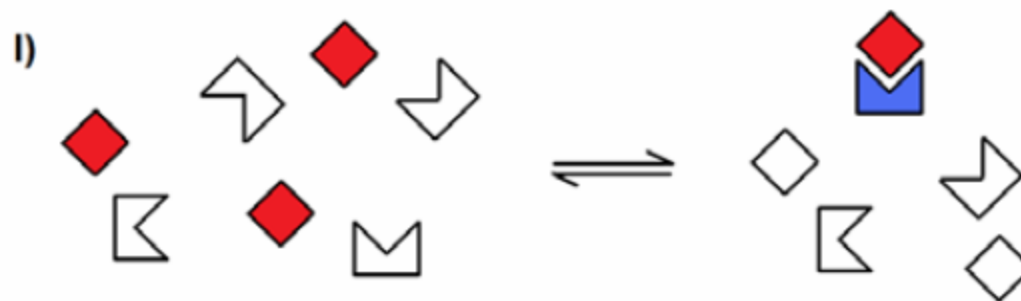
AFP



(R = C<sub>14</sub>H<sub>29</sub>)

Analyte (TNT)

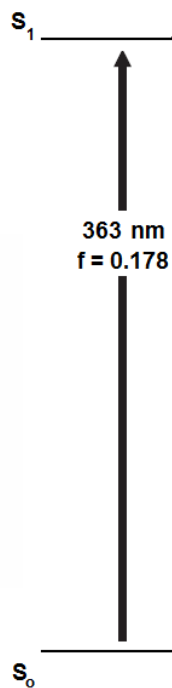
Receptor



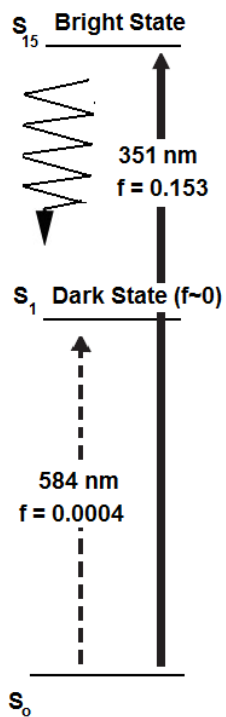


# Quenching

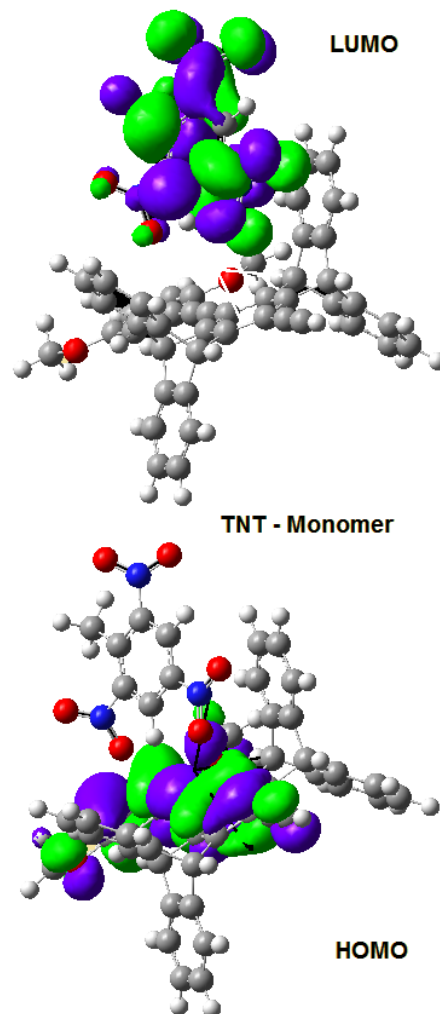
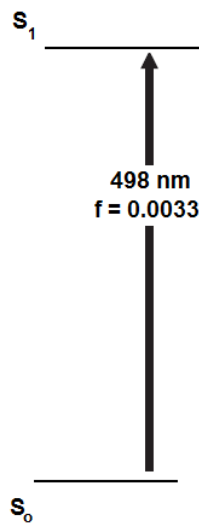
AFP Monomer



AFP Monomer - TNT Complex



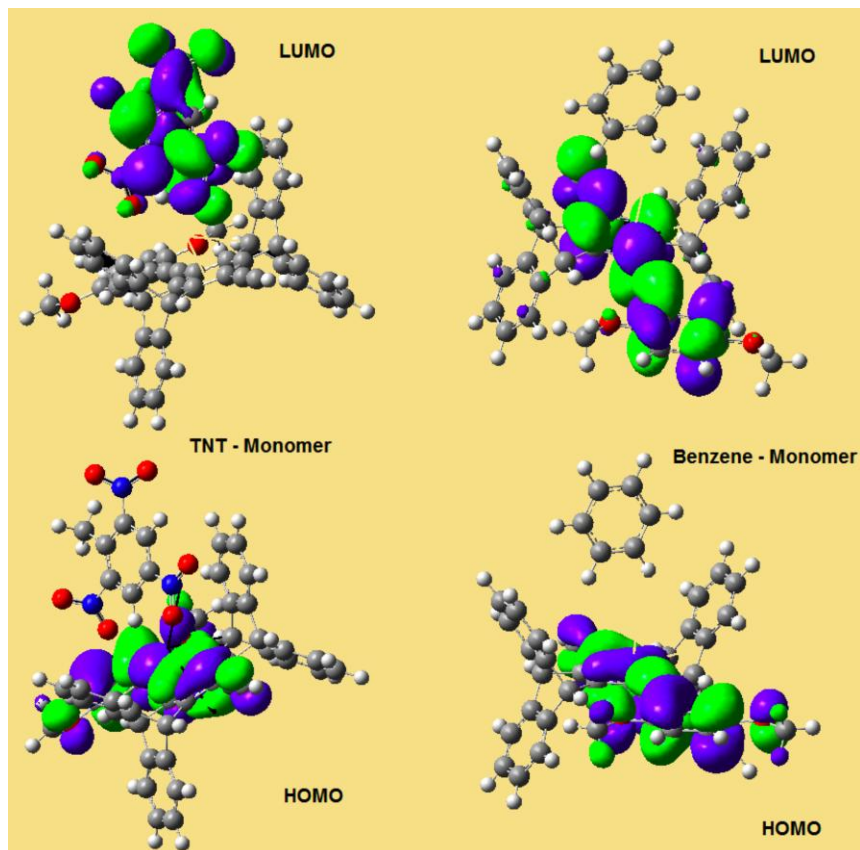
TNT



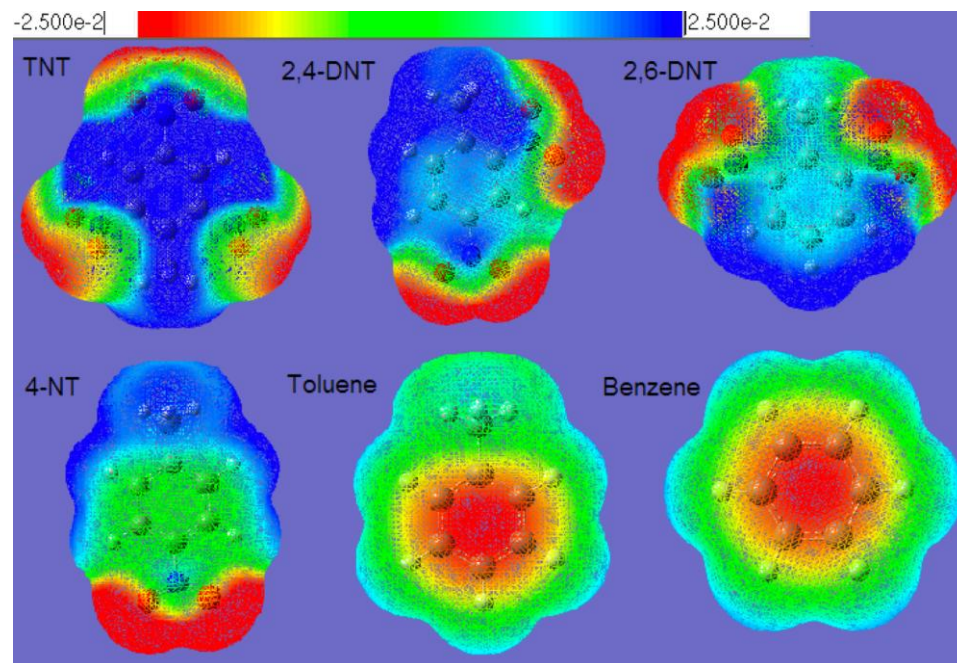


# Time Dependent DFT study of IED Sensing Mechanisms

B. Arman, H. Fan, T. Cagin, J. Chem. Phys. Submitted.



TNT	-3.92990
2,4-DNT	-3.40907
2,6-DNT	-3.28362
4-NT	-2.79245
Toluene	-0.34776
Benzene	-0.39294

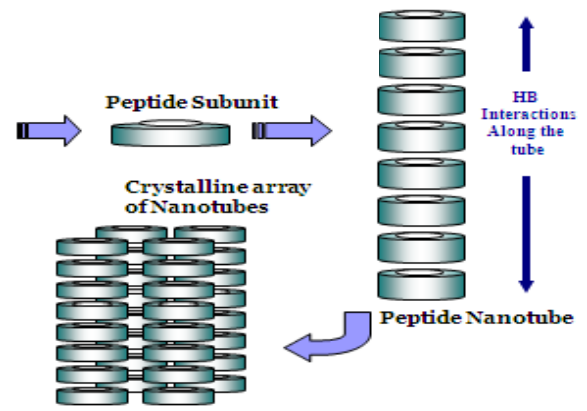
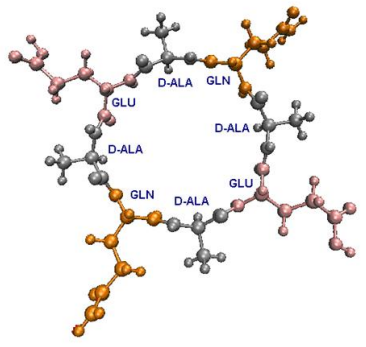


Analyte Name	Analyte Homo Lumo-Gap (eV)	Analyte-Monomer Homo-Lumo Gap (eV)
TNT	4.880	2.029
2,4 DNT	5.004	2.360
2,6 DNT	4.948	2.390
4 NT	4.862	2.927
Toluene	6.328	3.966
Benzene	6.602	3.967

# Stability and Optimization

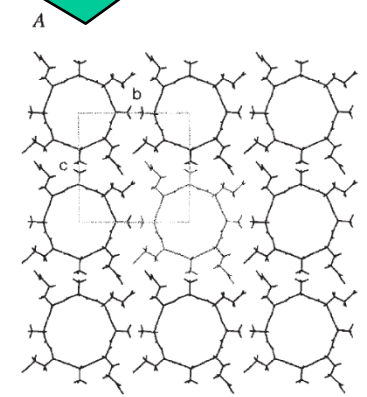
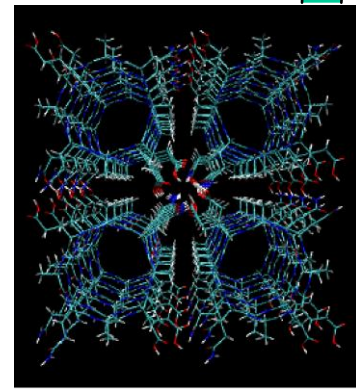


Basic Peptide Subunit

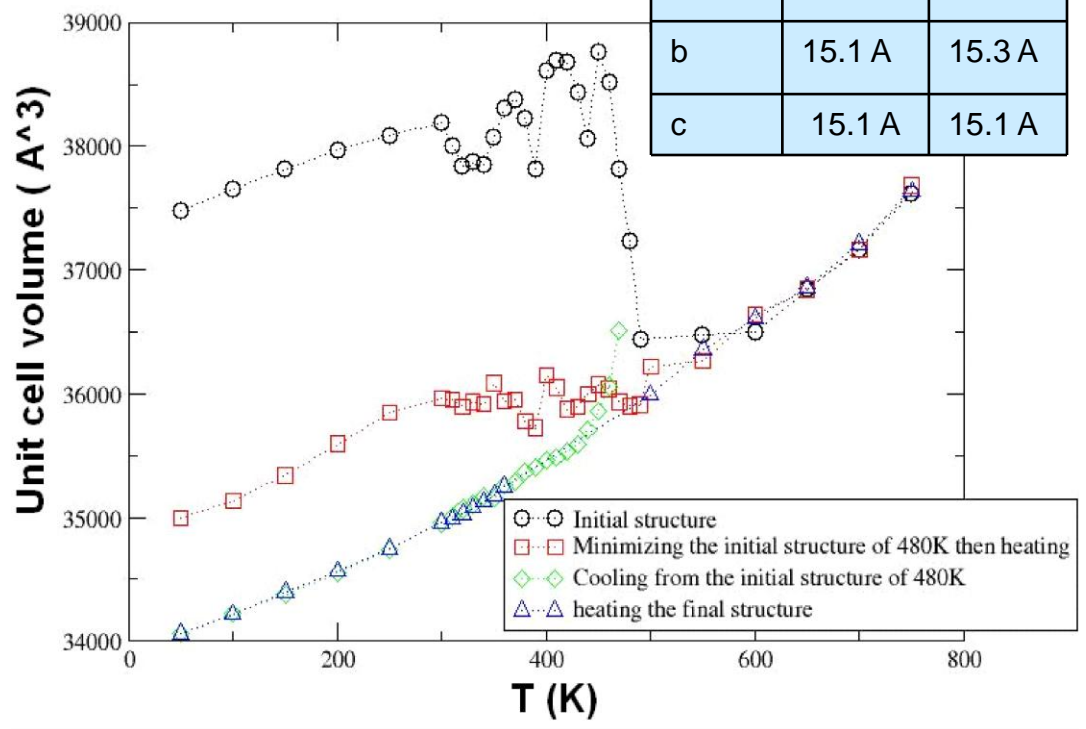
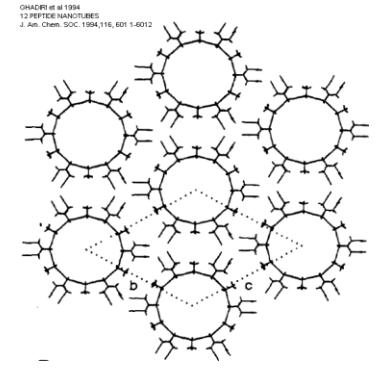
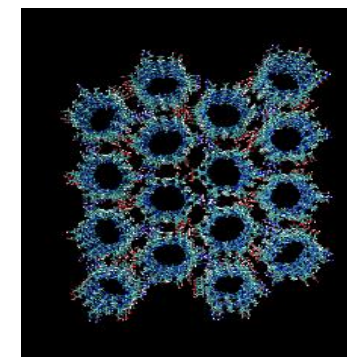


Crystalline nanotubes

8-Peptide System



12-Peptide System







# Transport Properties



Diffusion of water in Peptide Nanotubes is faster compared with equivalent diameters of CNTS.

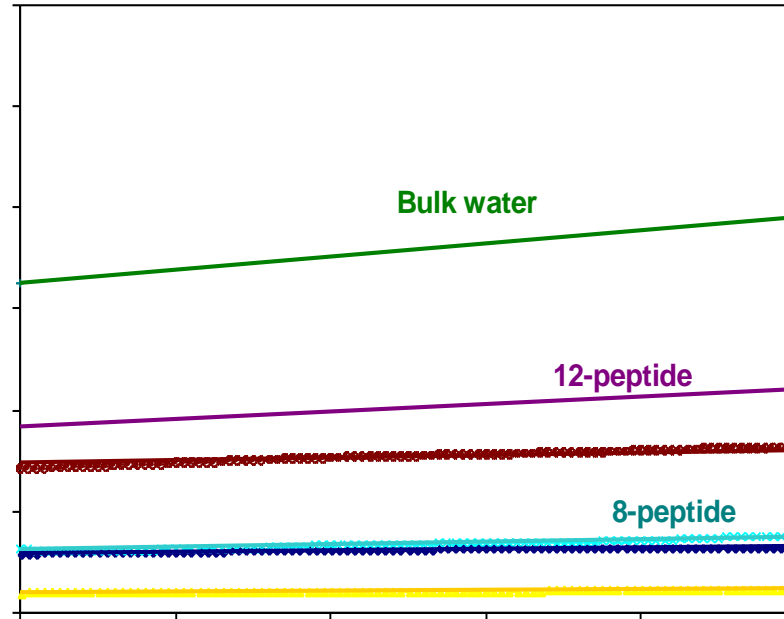
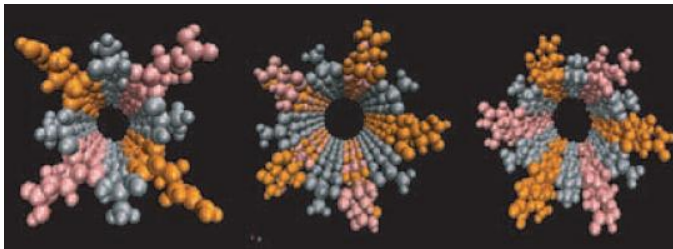
## Self Diffusion Coefficient

$$\langle (x(t + \Delta t) - x(t))^2 \rangle = 6 * D * t$$

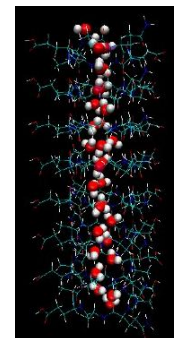
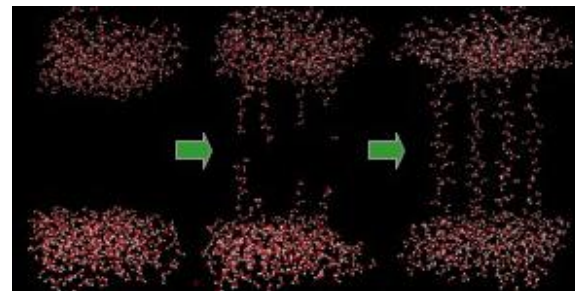
*Einstein's Relationship*

### DETAILS

From the analysis of curves of mean square displacement along axial direction.



System	Diameter (Å)	Diff. coeff. (calc) cm <sup>2</sup> /s (1 * 10 <sup>5</sup> )
Bulk water	—	2.17
12-peptide	14	1.23
(15,15) CNT	15	0.41
8-peptide	8.8	0.41
(9,9) CNT	8.6	0.25
(8,8) CNT	7.2	0.13

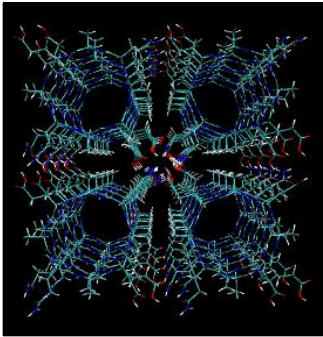
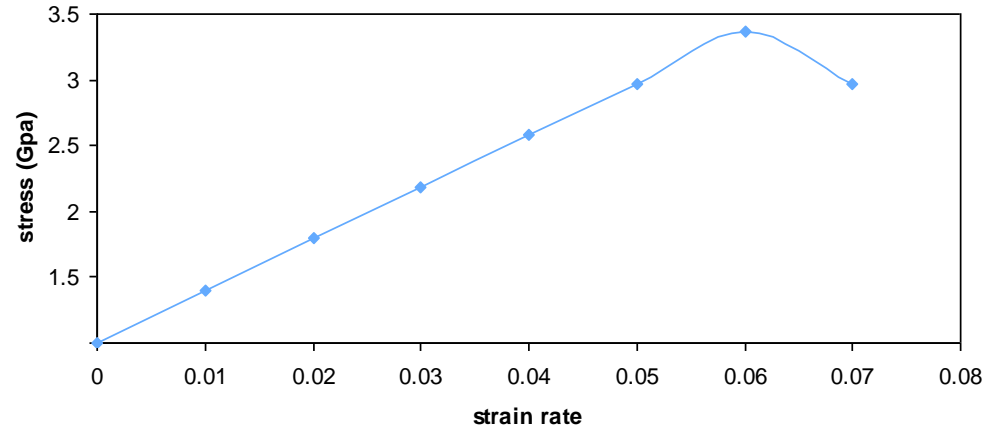




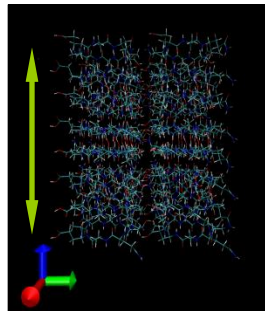
# Mechanical Properties



Stress-Strain



SIDE CHAIN- SIDE CHAIN  
INTERTUBULAR HYDROPHOBIC  
INTERACTIONS



HYDROGEN BONDING  
INTERACTION ALONG THE  
NANOTUBES

$$\Delta E = \frac{V_0}{2} \sum C_{IJ} \eta_I \eta_J + \frac{V_0}{6} \sum C_{IJK} \eta_I \eta_J \eta_K$$

Anisotropic Isothermal  
Elastic Constants

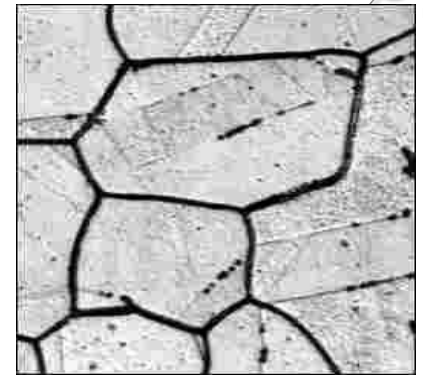
Cij	value(Gpa)	Cij	value(Gpa)
C11	8.09	C66	0.77
C22	10.16	C12	6.56
C33	19.65	C13	9.56
C44	1.23	C14	0.57
C55	1.23	C23	9.59

*Experimental Young Modulus reported for Peptide Nanotubes :19GPa. Self-Assembled Peptide Nanotubes Are Uniquely Rigid Bioinspired Supramolecular Structures. Nano Lett., 2005, 5 (7), pp 1343-1346*



# Stress-Corrosion Cracking (SCC) in Fe

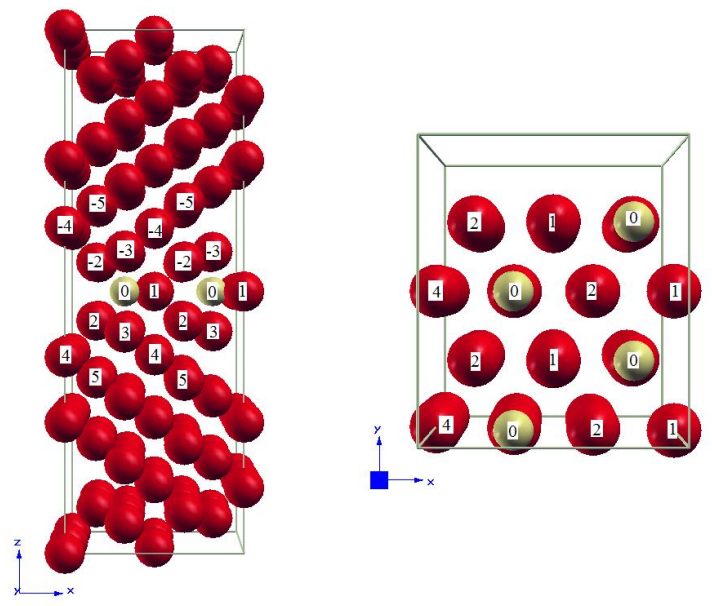
- Concerns vast range of application
- Combined influence of stress & corrosive environment
- SCC is proved to be connected to GB
  - introduction of impurity element
  - giving no sign of warnings



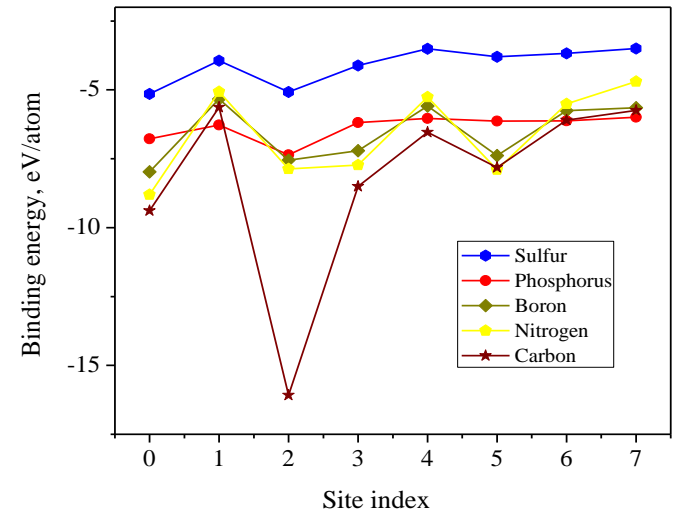
(Source: Corrosion testing lab)

## Binding energy

$$E_b N_I = E_{tot}^{GB}(N_{Fe}, N_I) - N_I E_I - E_{tot}^{GB}(N_{Fe}^0) - \frac{N_{Fe} - N_{Fe}^0}{N_{Fe}^0} E_{tot}^{bulk}(N_{Fe}^0)$$



Σ3 (111) grain boundary - 96 Fe atoms



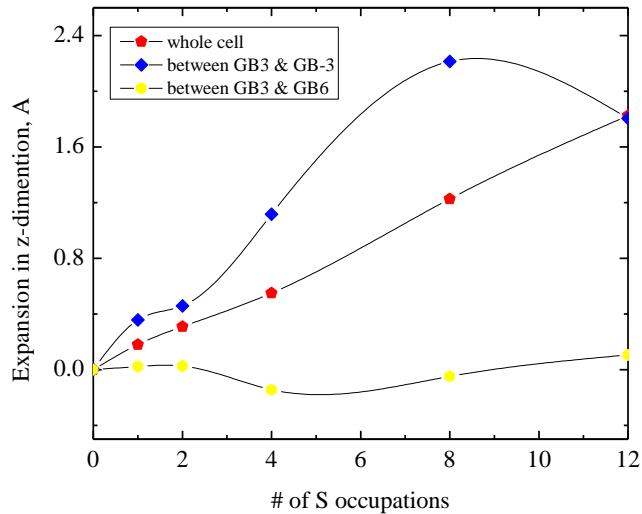
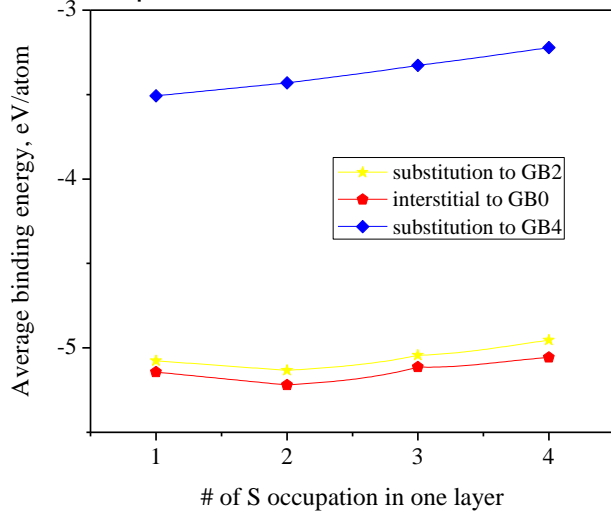
- Carbon ties strongly
- GB0 & GB+/-2 are favorite sites: geometry other than chemistry



# Behavior of Sulfur segregation

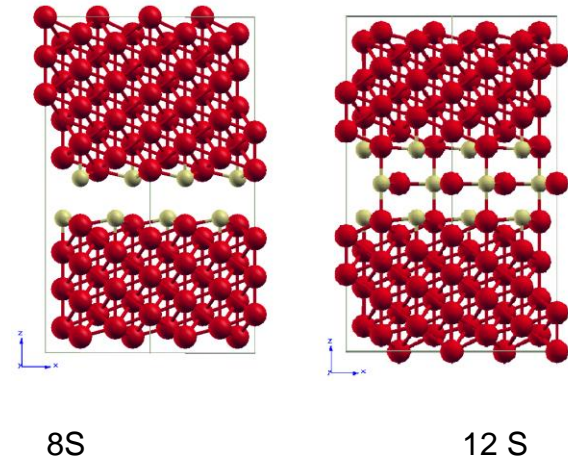


Fig. Average binding energy of Sulfur as function of layer occupation



Tab. Behavior of GB cell under S attachment  
 a, b, c - size of GB cell in x, y, and z dimaentions, respectively  
 d - distance between GB3 & GB-3

layer	# of occ.	a, Å	b, Å	c, Å	d, Å	- Eb/S, eV
clean cell	0	6.92	7.99	20.29	3.15	--
GB0	1	6.92	8.00	20.47	3.51	-5.14
	2	6.92	8.03	20.60	3.61	-5.22
	3	6.96	8.00	20.75	3.68	-5.11
	4	6.96	8.02	20.84	4.27	-5.06
GB0 & GB2	8	6.89	7.96	22.12	5.36	-4.86
GB0 & GB2 & GB-2	12	6.94	8.02	22.12	4.95	-4.46



- z-expansion due to GB separation
- S atoms expose repulsive forces
- interactions around GB broken



# Behavior of GB cell under P, N, C and B attachment



Elements	# of occ.	$\Delta a, \text{\AA}$	$\Delta b, \text{\AA}$	$\Delta c, \text{\AA}$	$\Delta d, \text{\AA}$	-Eb/S, eV
<b>P</b>	1	0.00	0.00	0.19	0.34	-6.78
	2	0.00	0.00	0.37	0.45	-7.46
	4	0.01	0.02	0.60	1.15	-7.41
	8	-0.04	-0.06	0.80	1.14	-6.22
	12	0.10	-0.14	1.13	1.25	-5.91
<b>N</b>	1	0.03	-0.01	-0.02	0.29	-8.80
	2	0.02	0.05	-0.03	0.39	-8.95
	4	0.09	-0.01	0.04	0.41	-7.92
	8	0.16	-0.10	-0.08	0.17	-8.15
	12	0.22	-0.21	-0.30	-0.25	-7.74
<b>C</b>	1	0.02	0.00	-0.01	0.25	-9.38
	2	0.02	0.04	-0.02	0.31	-9.40
	4	-0.02	-0.01	0.21	0.56	-8.85
	8	-0.05	-0.05	0.13	0.39	-7.92
	12	-0.19	-0.23	0.53	0.33	-7.42
<b>B</b>	1	-0.01	-0.01	0.08	0.25	-7.98
	2	-0.01	-0.02	0.18	0.27	-7.96
	4	-0.04	-0.04	0.40	0.81	-7.94
	8	-0.07	-0.10	0.20	0.61	-6.98
	12	-0.16	-0.18	0.50	0.52	-6.39

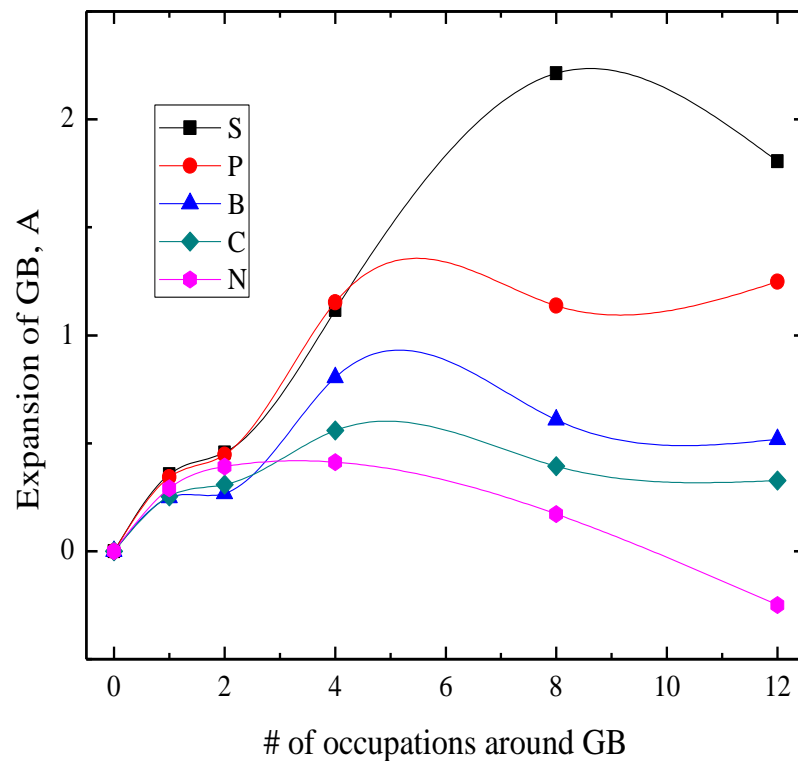


Fig. Comparative separation of Fe  $\Sigma 3$  (111) GB under the attack of different impurity atoms (S, P, N, C, B)

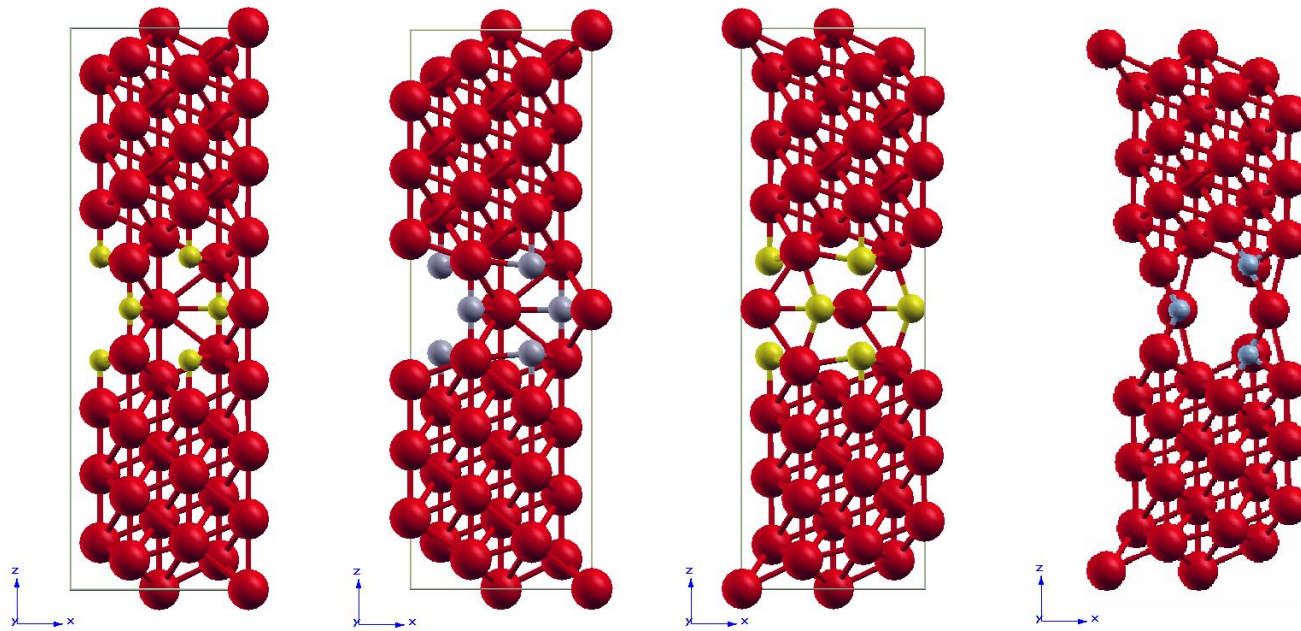


Fig. Behavior of Fe  $\Sigma 3$  (111) GB due to the precipitation of C, B, P and N

- The same binding tendency to a specific locations at GB
- Little interactions from impurity particles on the same layer
- S & P causes the separation of GB, which may initiate cracks
- B & C have little effects on GB mechanical properties
- N weakens the GB structure through formations of cavities and voids



- **First Principles DFT+U studies on (Ce,Th) O<sub>2</sub> alloys**
  - Structure, Mechanics, Dynamics, Alloying of CeO<sub>2</sub> and ThO<sub>2</sub>
  - C. Sevik, T. Cagin, “Mechanical and electronic properties of CeO<sub>2</sub>, ThO<sub>2</sub>, and (Ce, Th)O<sub>2</sub> alloys” submitted to Phys Rev B. (2009)

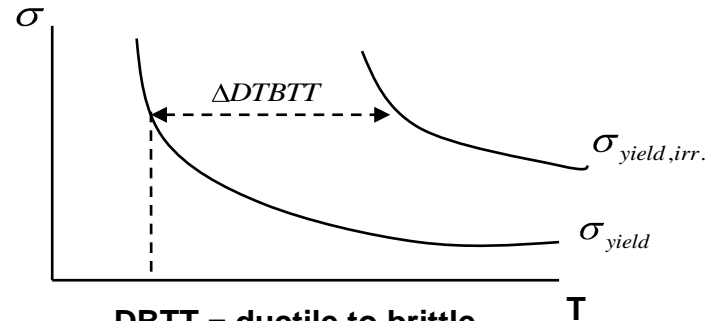
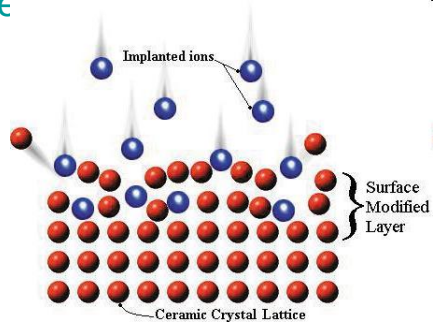
Calculated lattice parameters, mechanical properties for C<sub>x</sub>Th<sub>1-x</sub>O<sub>16</sub>.

	a <sub>0</sub>	B <sub>0</sub>	C <sub>11</sub>	C <sub>12</sub>	C <sub>44</sub>	Alloy
LSDA+U	5.571	214	379	131	104	Ce <sub>1</sub> Th <sub>7</sub> O <sub>16</sub>
LDA	5.507	216	386	131	95	
LSDA+U	5.548	215	382	132	101	Ce <sub>2</sub> Th <sub>6</sub> O <sub>16</sub>
LDA	5.488	215	385	130	92	
LSDA+U	5.500	213	382	129	96	Ce <sub>4</sub> Th <sub>4</sub> O <sub>16</sub>
LDA	5.448	210	379	126	87	
LSDA+U	5.450	215	386	130	88	Ce <sub>6</sub> Th <sub>2</sub> O <sub>16</sub>
LDA	5.405	209	377	125	79	
LSDA+U	5.425	216	388	130	85	Ce <sub>7</sub> Th <sub>1</sub> O <sub>16</sub>
LDA	5.383	208	376	124	76	



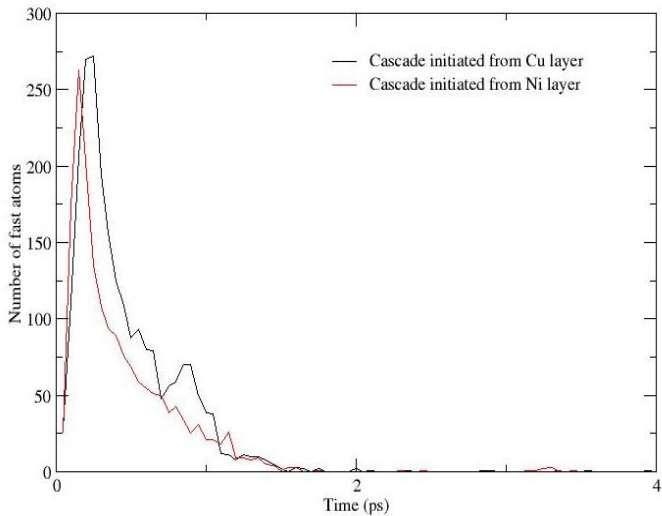
## High speed particle impact on atomic scale

- Radiation damage, degradation and embitterment (nuclear material shields, space gadgets etc.)
- Ion implantation, deposition (semiconductor device production)
- Surface modification (surface hardening, corrosion resistance  $\epsilon^{4-5}$ )

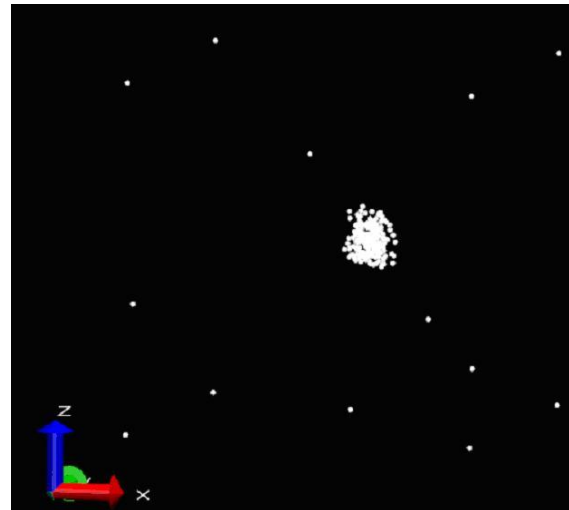


DBTT = ductile to brittle transformation temperature

## Molecular Dynamics simulations of irradiation process



Thermal spike and following thermalization in Cu-Ni superlattice



Simulation of microstructure evolution under irradiation in Cu-Ni superlattice





# Magnetic Shape Memory Alloys

## -Ni<sub>2</sub>MnIn

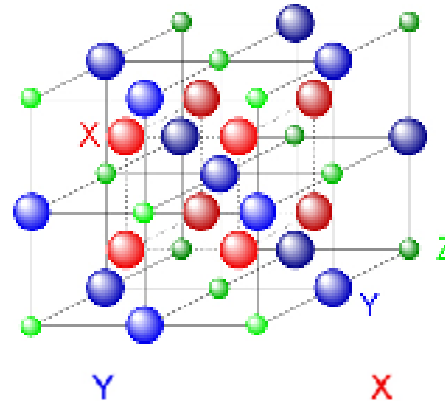
- **Heusler alloy** structure
  - L21 in austenite phase
- Ferromagnetic due to separation of magnetic moments residing on Y atoms
- Ni<sub>2</sub>MnGa most extensively studied, with reported recoverable strains  $\approx 10\%$  in the martensite phase



University of Alberta



USAF Aircraft Pictures - <http://sun.vmi.edu/hall/afpics.htm>



H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pd	Bi	Po	At	Rn
Fr	Ra																
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	
		Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

[http://www.riken.jp/lab-www/nanomag/research/heusler\\_e.html](http://www.riken.jp/lab-www/nanomag/research/heusler_e.html)





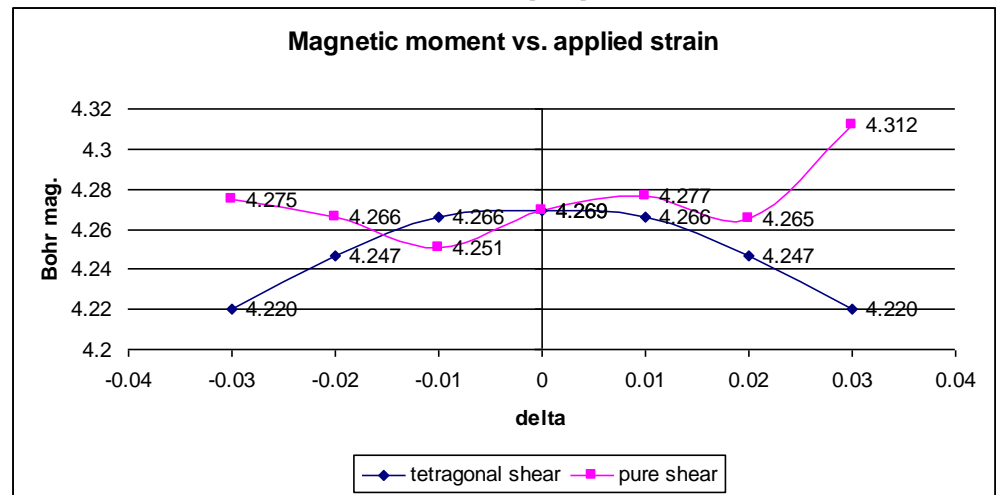
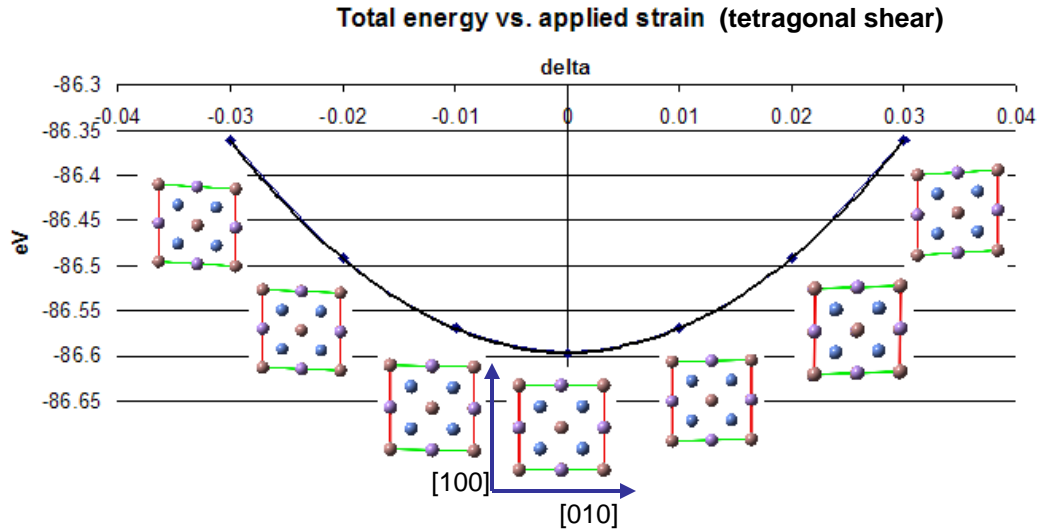
# Magnetostructural Coupling in Ni<sub>2</sub>MnIn

We apply volume-conserving strains to determine the magneto-mechanical response:  
-tetragonal shear

$$\varepsilon = \begin{bmatrix} \delta & 0 & 0 \\ 0 & \delta & 0 \\ 0 & 0 & \frac{-\delta^2 - 2\delta}{(\delta + 1)^2} \end{bmatrix}$$

-pure shear

$$\varepsilon = \begin{bmatrix} 0 & \delta & 0 \\ \delta & 0 & 0 \\ 0 & 0 & \frac{\delta^2}{1 - \delta^2} \end{bmatrix}$$

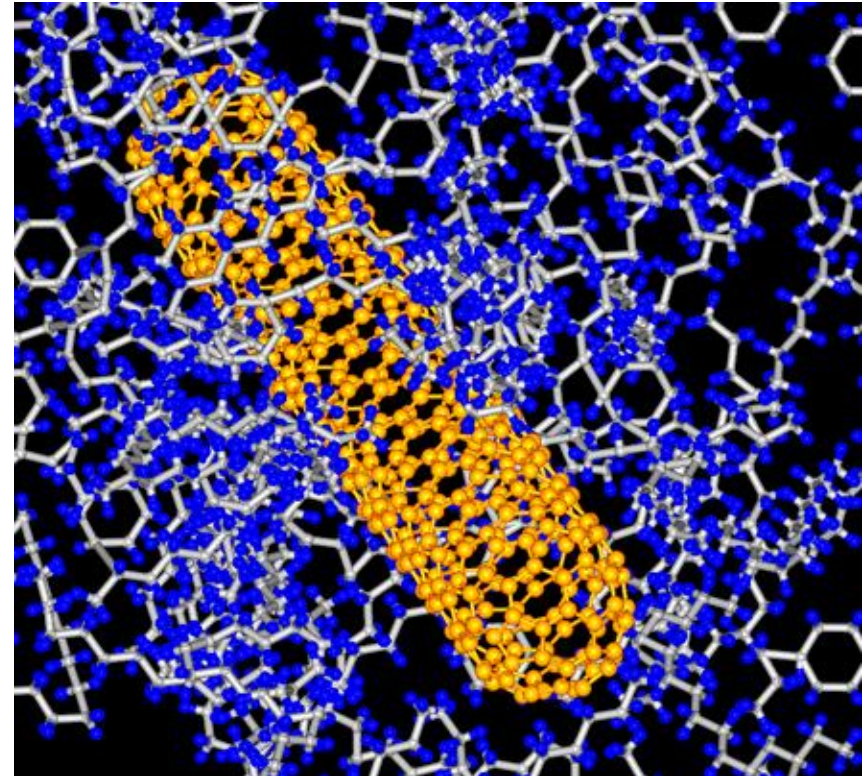


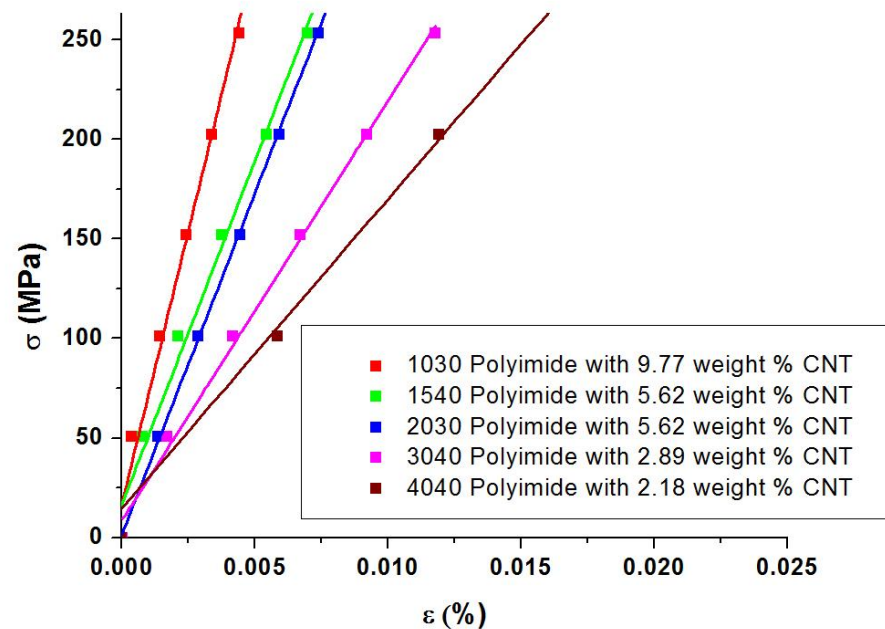
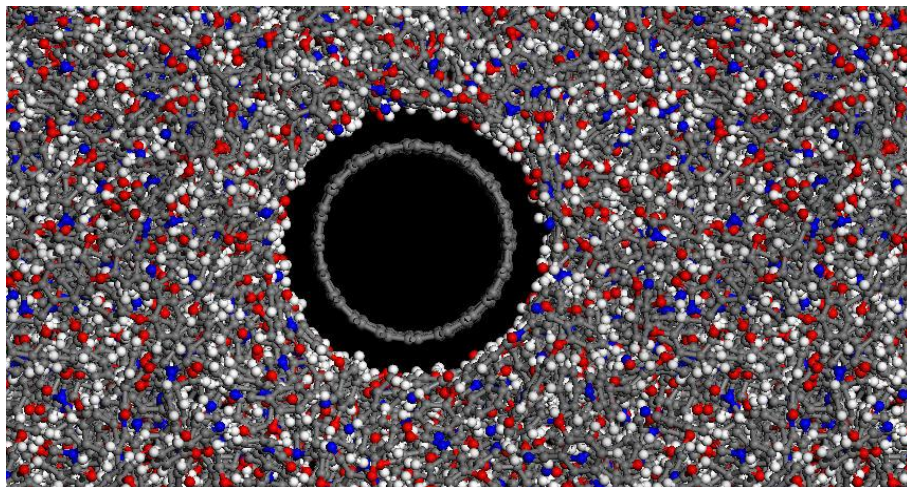
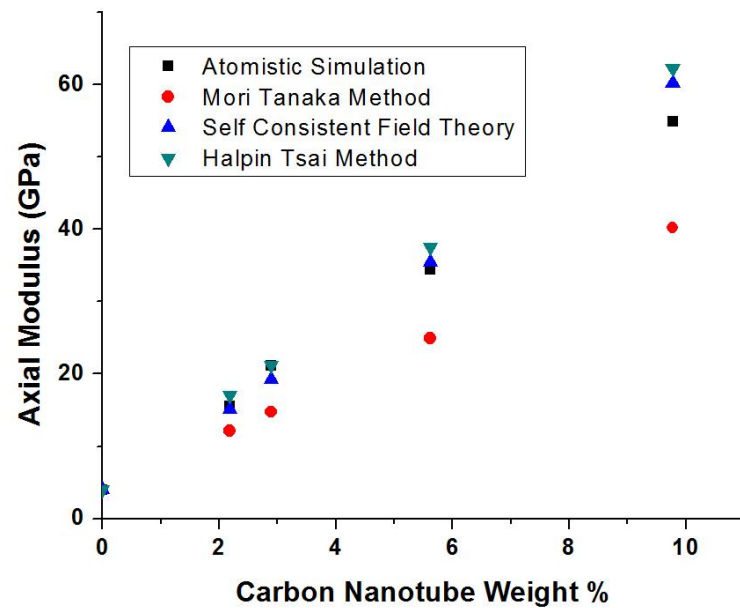
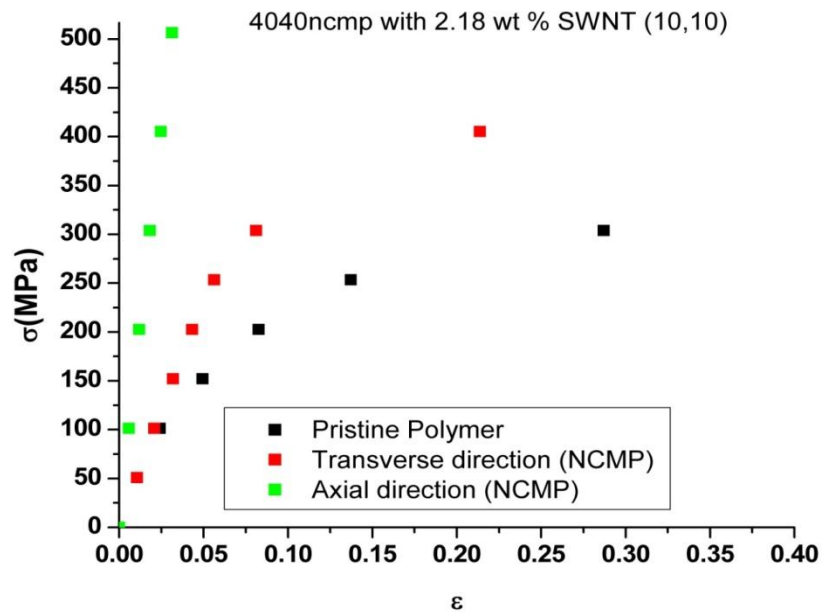


# Polyimide-nanotube composites for electro-active materials

A. CHAKRABARTY, T. CAGIN, CMC 3, 167 (2008); MMM (2008)

- ( $\beta$  – CN)APB/ODPA Polyimide
- Piezoelectric polyimide
- Exceptional thermal, mechanical, and dielectric properties
- Amorphous in nature
- Potential use in high temperature application







# Acknowledgements



Financial Support: NSF, DARPA, ONR, ARO, DOE, & AFRL

NSF (ITR-ASE: stress corrosion)  
NSF (IGERT): nanofluidics, SMA, CPNTs  
NSF: fire retardant PNC's  
DARPA (PROM: FE and TE materials)  
ONR (Energetic Materials)  
ONR (H-Pd under extreme conditions)  
ARO (Energetic Materials)  
AFRL (Thermo electrics)  
AFRL (IED Sensing)  
DOE (Nuclear Fuels)  
DOE (Multiscale Modeling)  
CONACyT (Domain walls in FE devices)  
CONACyT (Dielectric Gate Stacks)  
TAMU (Transport in bio-nano systems)  
PIIF (H-storage systems)  
TUBITAK (Si-nanocrystals)  
TUBITAK (MSMA's)  
PMMA thin film electronics

ARMAN, HASKINS  
CHAKRABARTY, KINACI, SEVIK  
PHAM, SHIV, OJEDA, CAGIN  
KAMANI, LIZAROZU  
BISWAS, CARVAJAL, WILLIAMS, NJOREGE



TAMU Super Computing Facility